



JOHANNES GUTENBERG
UNIVERSITÄT MAINZ



www.superheavies.de
c.e.duellmann@gsi.de

Quest for Superheavy Elements

Christoph E. Düllmann

Johannes Gutenberg-Universität, Mainz

GSI Helmholtzzentrum für Schwerionenforschung GmbH, Darmstadt

Helmholtz Institut Mainz



Ecole Joliot-Curie "Nucleus through the looking glass – High intensity stable and ISOL beam frontier"

La Villa Clythia – Fréjus – France – September 30-October 05, 2012

What's on the menu this week?

Lesson 1:

- Discovery of the transuranium elements: $Z=93 - 112$
- Stability of superheavy elements I

Lesson 2:

- Discovery of the transuranium elements: $Z=113- \dots$
- Stability of superheavy elements II

Lesson 3:

- Reactions: synthesis of SHE
- Search for new elements at GSI

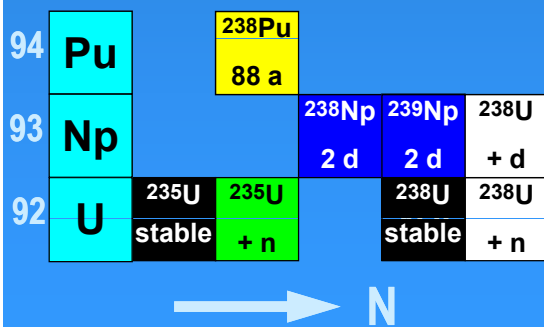
Introduction: looking back a few decades...

The Periodic Table 1939

1 H																	2 He	
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
55 Cs	56 Ba	57-71 La - Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
87 Fr	88 Ra	89 Ac	90 Th	91 Pa	92 U	(93)	(94)	(95)	(96)	(97)	(98)	(99)	(100)					
		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu		

Np: Milking of U-daughter; not Re-like

Pu: not Os-like

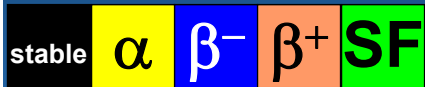


	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17
	Sc	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I
	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At
	Ac	Th	Pa	U	Np	Pu									
	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

Np: Milking of U-daughter; not Re-like

Pu: not Os-like \Rightarrow Uranide-concept

E95 attempts fail



95	Am	²³⁹ Am	²⁴⁰ Am	²³⁹ Pu	
		?	?	+ d	
94	Pu	²³⁸ Pu	²³⁹ Pu	²⁴⁰ Pu	²⁴¹ Pu
		88 a	2e5 a		14 a
93	Np		²³⁸ Np	²³⁹ Np	²³⁸ U
			2 d	2 d	+ d
92	U	²³⁵ U	²³⁵ U	²³⁸ U	²³⁸ U
		stable	+ n	stable	+ n



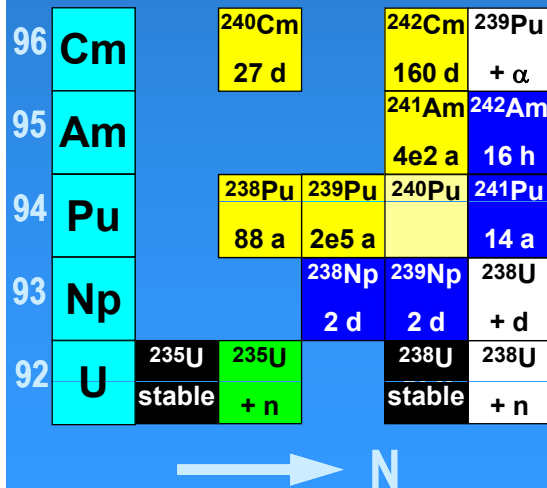
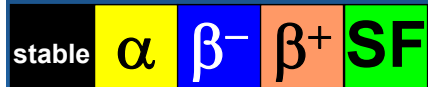
	4	5	6	7	8	9	10	11	12	13	14	15	16	17	
	Sc	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I
	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At
	Ac	Th	Pa		U	Np	Pu	95							
	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

Np: Milking of U-daughter; not Re-like

Pu: not Os-like \Rightarrow Uranide-concept

E95 attempts fail \Rightarrow Actinide concept

Am+Cm thanks to An concept



	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17
	Sc	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I
	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At
	Ac														
	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
	Ac	Th	Pa	U	Np	Pu	Am	Cm							

Np: Milking of U-daughter; no
 Pu: not Os-like \Rightarrow Uranide-con
 E95 attempts fail \Rightarrow Actinide c
 Am+Cm thanks to An concept
 Bk+heavier: Chromatographic

Z
 \uparrow

stable α β^- β^+ SF

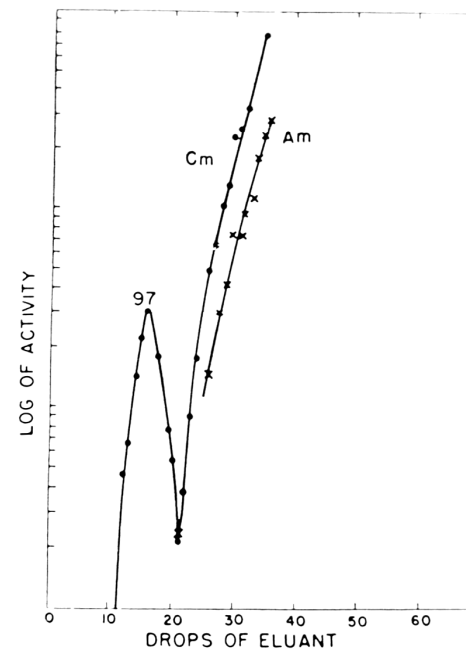


Fig. 5.3. Original elution data corresponding to the discovery of berkelium (^{243}Bk); S.G. Thompson, A. Ghiorso, and G.T. Seaborg, December 19, 1949; Dowex-50 eluted with citrate at 87°C .

97	Bk		^{243}Bk 5 h	^{241}Am + α
96	Cm	^{240}Cm 27 d	^{242}Cm 160 d	^{239}Pu + α
95	Am	^{239}Am ?	^{240}Am ?	^{241}Am 4e2 a
94	Pu	^{238}Pu 88 a	^{239}Pu 2e5 a	^{240}Pu 16 h
93	Np		^{238}Np 2 d	^{239}Np 2 d
92	U	^{235}U stable	^{235}U + n	^{238}U stable

\rightarrow N

	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17
	Sc	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I
	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At
	Ac														
	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk						

Np: Milking of U-daughter; not Re-like

Pu: not Os-like \Rightarrow Uranide-concept

E95 attempts fail \Rightarrow Actinide concept

Am+Cm thanks to An concept

Bk+heavier: Chromatographic separation of An

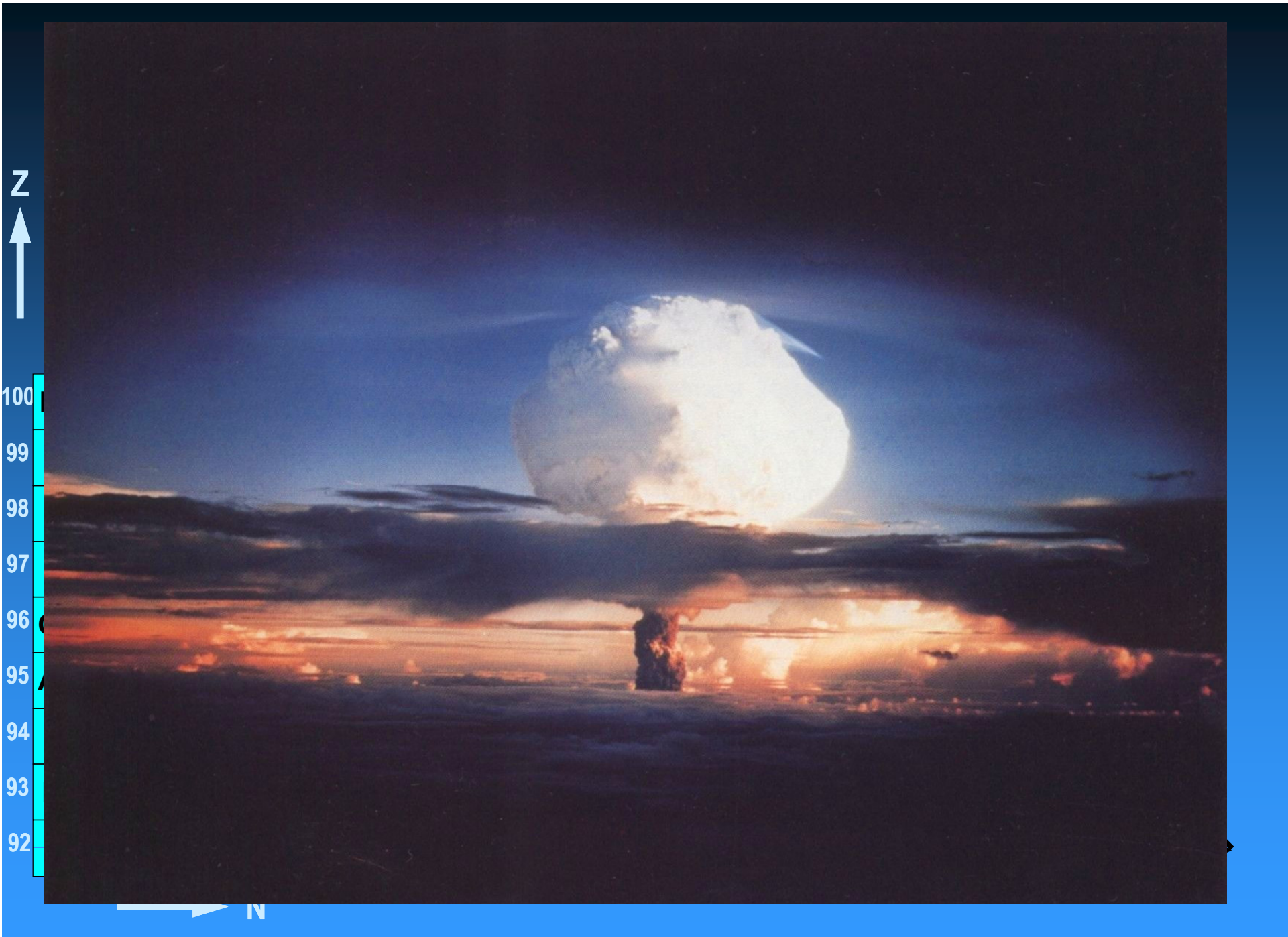


stable α β^- β^+ SF

98	Cf			²⁴⁵ Cf 43 m	²⁴² Cm + α
97	Bk			²⁴³ Bk 5 h	²⁴¹ Am + α
96	Cm	²⁴⁰ Cm 27 d		²⁴² Cm 160 d	²³⁹ Pu + α
95	Am	²³⁹ Am ?	²⁴⁰ Am ?	²⁴¹ Am 4e2 a	²⁴² Am 16 h
94	Pu	²³⁸ Pu 88 a	²³⁹ Pu 2e5 a	²⁴⁰ Pu	²⁴¹ Pu 14 a
93	Np		²³⁸ Np 2 d	²³⁹ Np 2 d	²³⁸ U + d
92	U	²³⁵ U stable	²³⁵ U + n	²³⁸ U stable	²³⁸ U + n



	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17
Sc	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	
La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	
Ac															
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf						



Np: Milking of U-daughter; not Re-like

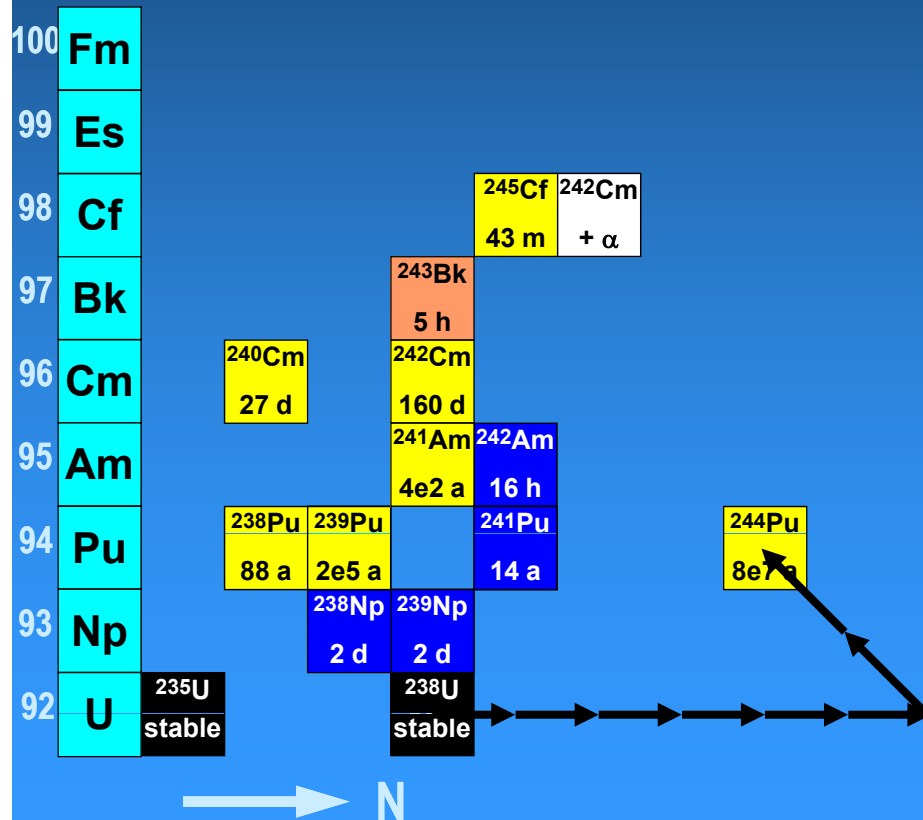
Pu: not Os-like \Rightarrow Uranide-concept

E95 attempts fail \Rightarrow Actinide concept

Am+Cm thanks to An concept

Bk+heavier: Chromatographic separation of An

Z
↑



Np: Milking of U-daughter; not Re-like

Pu: not Os-like \Rightarrow Uranide-concept

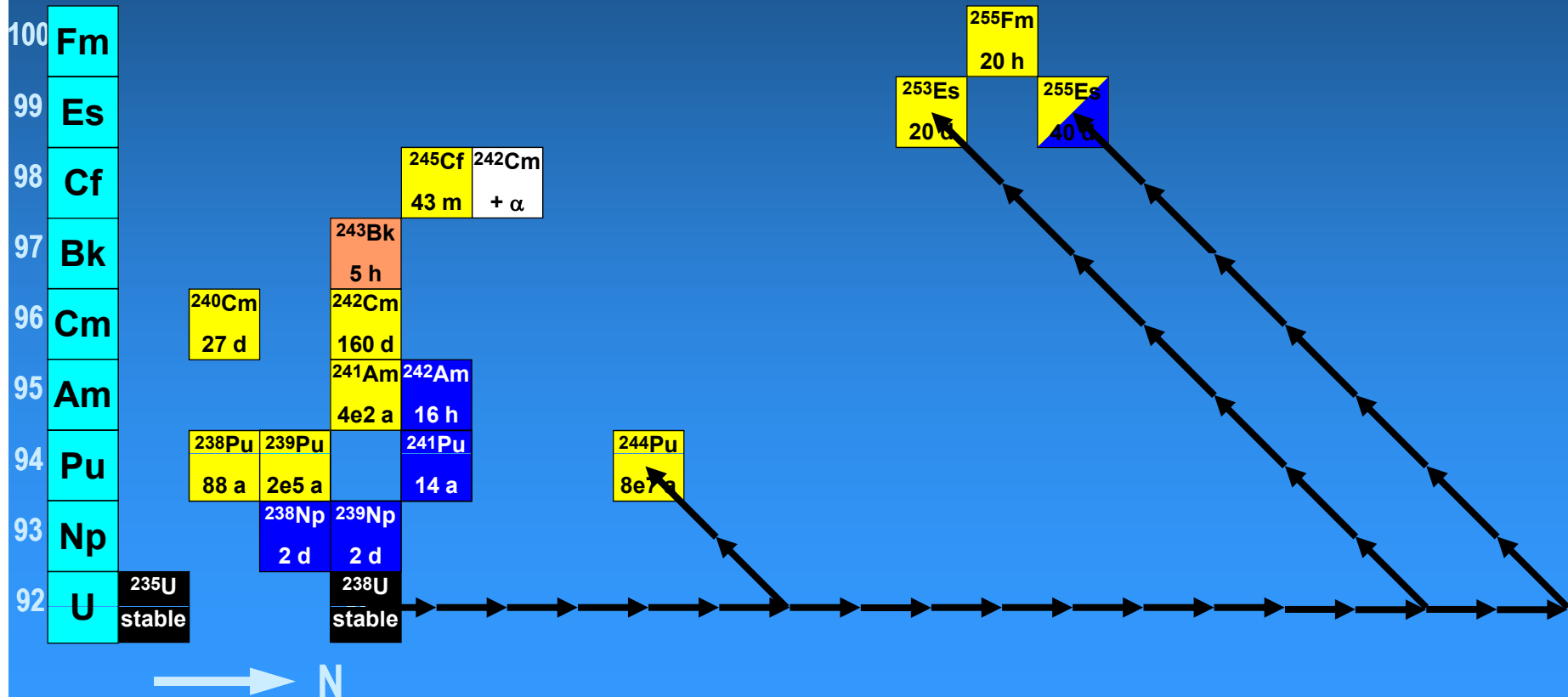
E95 attempts fail \Rightarrow Actinide concept

Am+Cm thanks to An concept

Bk+heavier: Chromatographic separation of An

Es/Fm: The elements from the bomb

Z
↑



Np: Milking of U-daughter; not Re-like

Pu: not Os-like \Rightarrow Uranide-concept

E95 attempts fail \Rightarrow Actinide concept

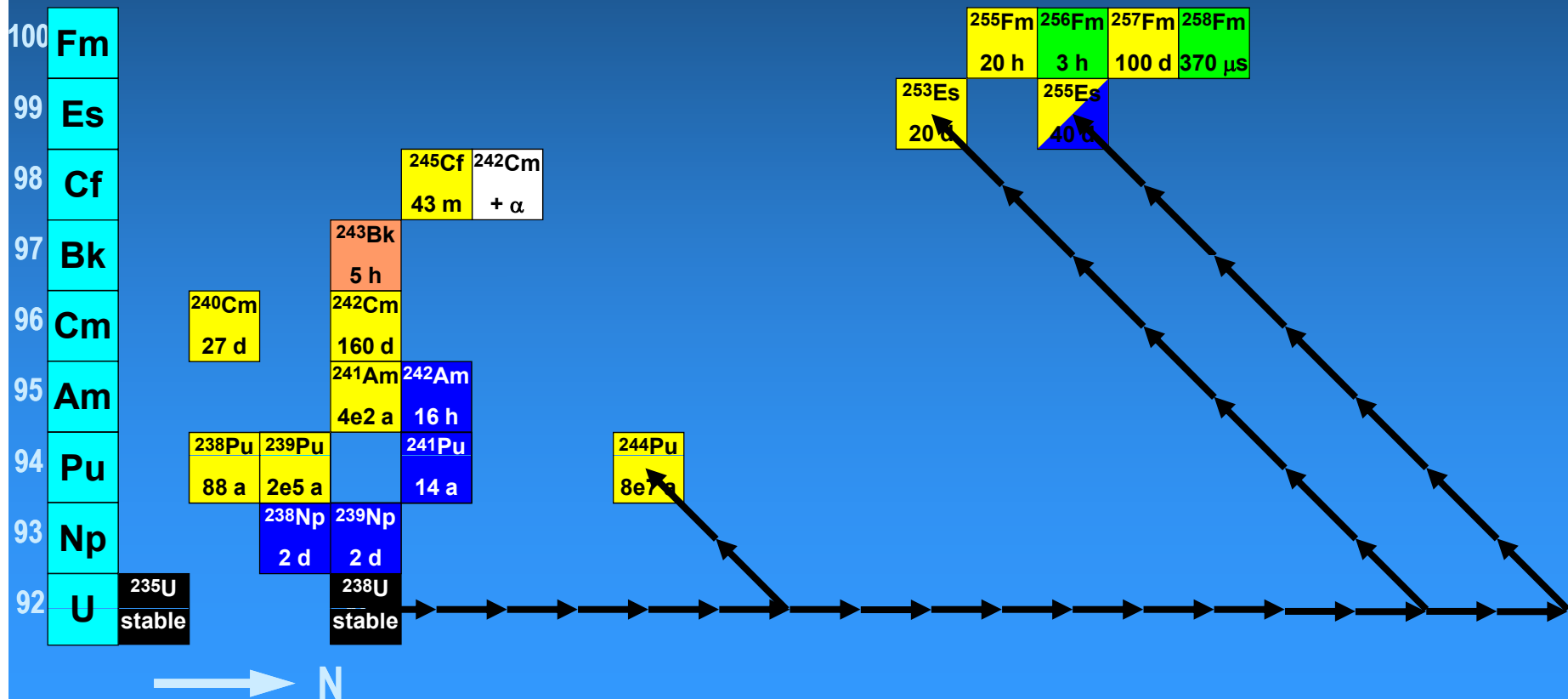
Am+Cm thanks to An concept

Bk+heavier: Chromatographic separation of An

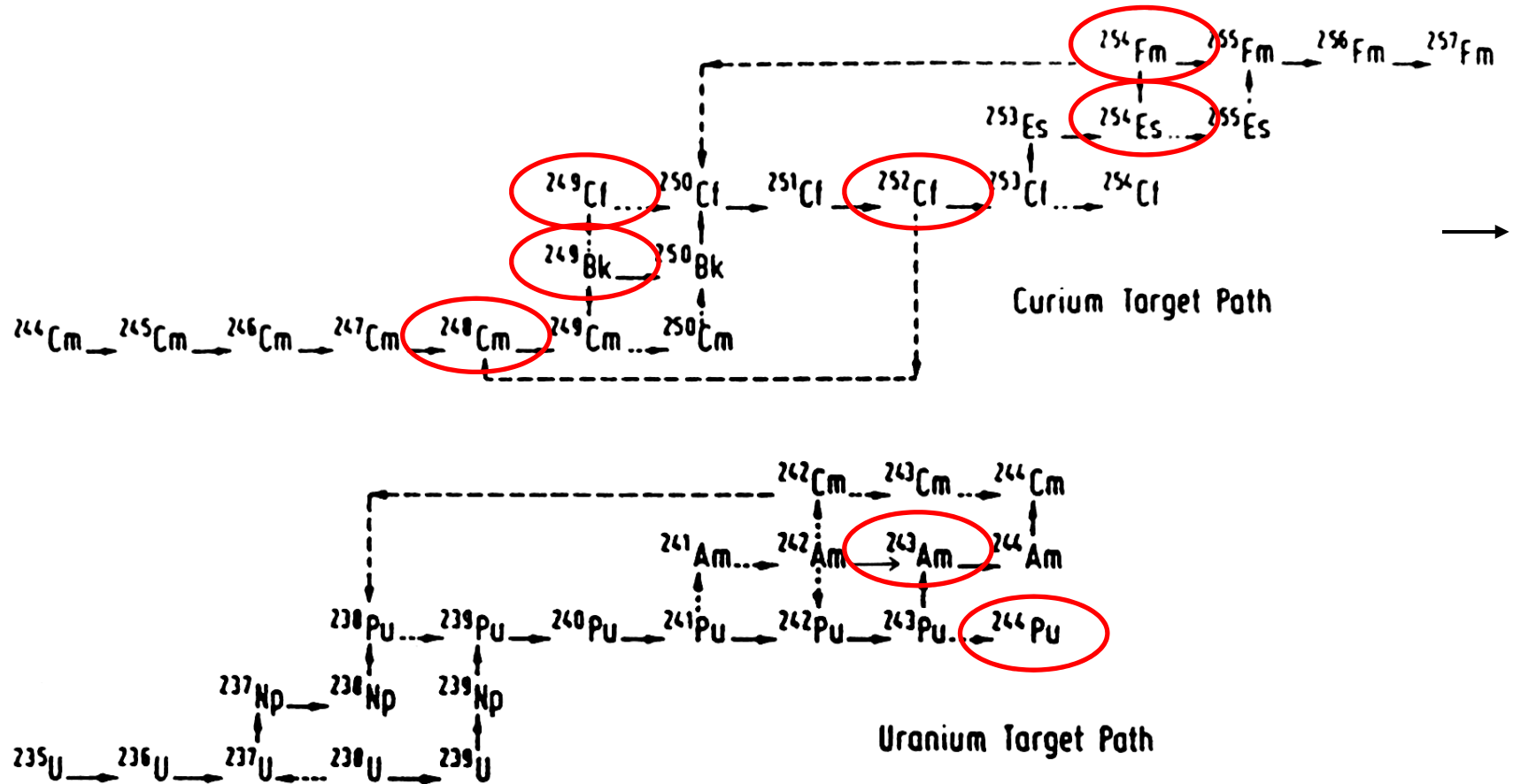
Es/Fm: The elements from the bomb

$T_{1/2}(\text{SF})$ of ^{258}Fm :

0.37 ms !!!



Production of transuranium nuclides in a high flux reactor



Fm is the heaviest element that can be produced in a reactor

Lessons learned

- Elements up to Fm can be produced in a reactor, heavier ones not
- Reactor-produced long-lived nuclides: targets for accelerator based experiments. Heaviest targets: ${}_{96}^{248}\text{Cm}$, ${}_{97}^{249}\text{Bk}$, ${}_{98}^{249}\text{Cf}$ (${}_{99}^{253}\text{Es}$)
- Nuclear properties can change dramatically by adding/removing one single nucleon
- Correct assumptions of the structure of the Periodic Table needed for chemical experiments

Map of the Nuclear Landscape

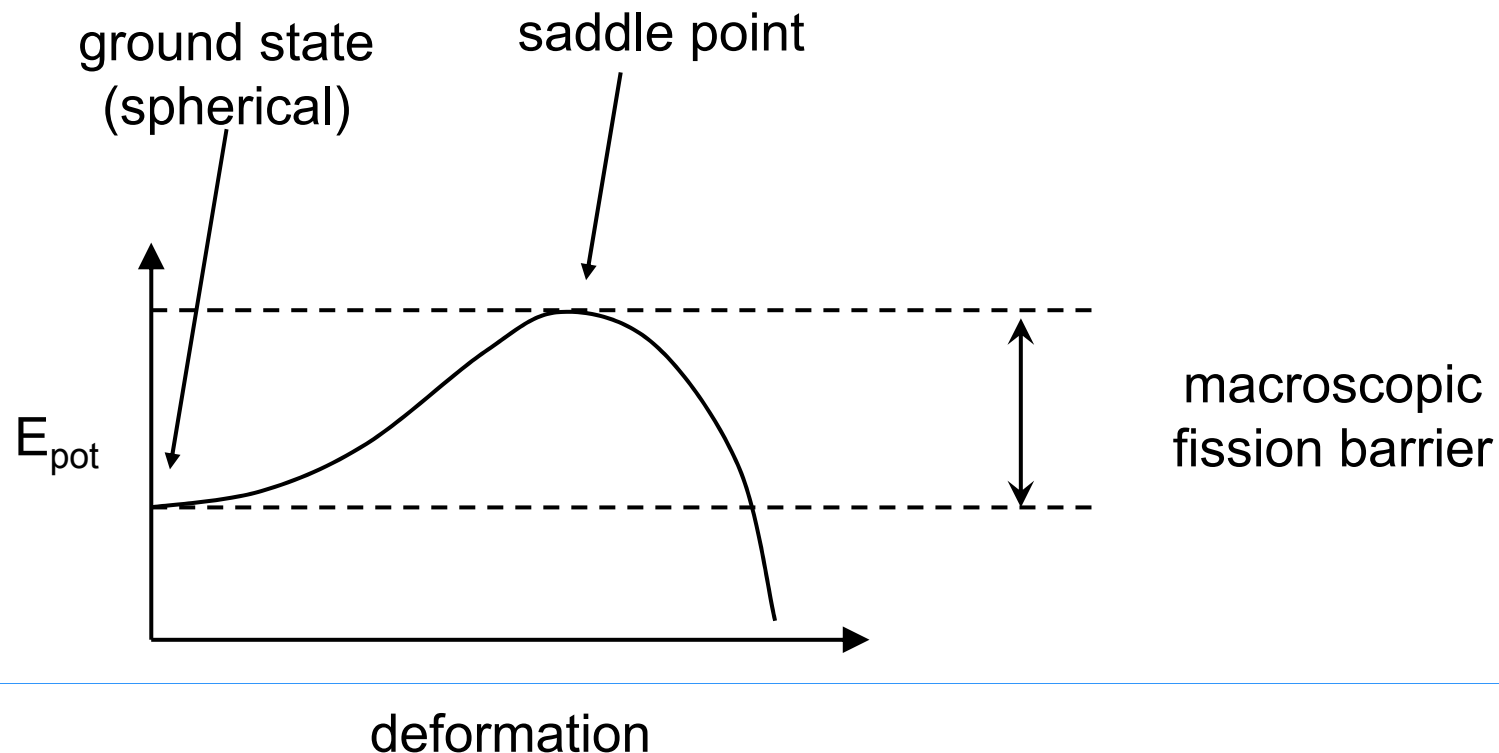


The Fission Barrier

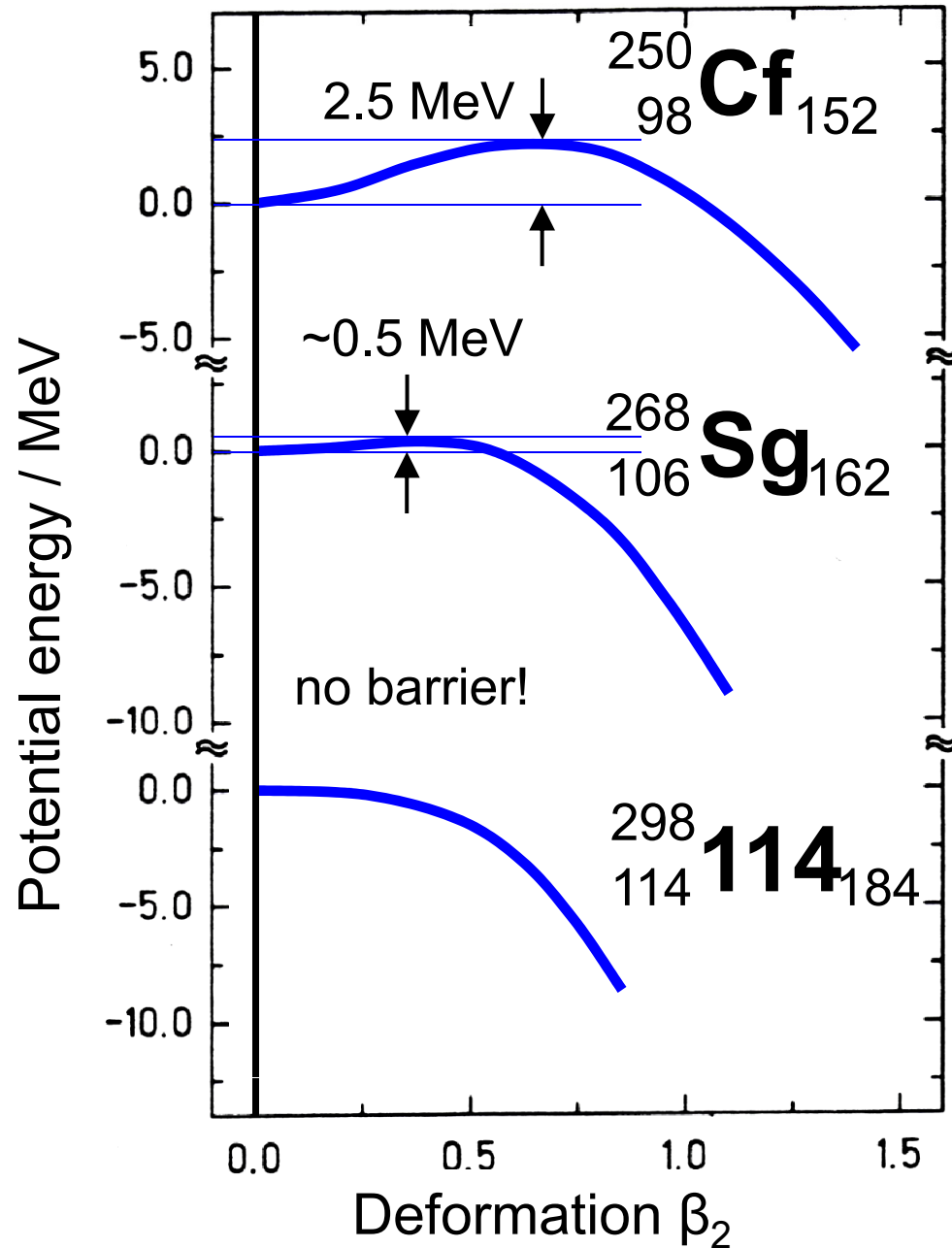
Liquid drop model includes two deformation-dependent terms:

The **Coulomb energie** decreases with increasing deformation due to the larger average proton-proton distance

The **surface energy** increases with increasing deformation as the sphere is the body with minimum surface for a given volume

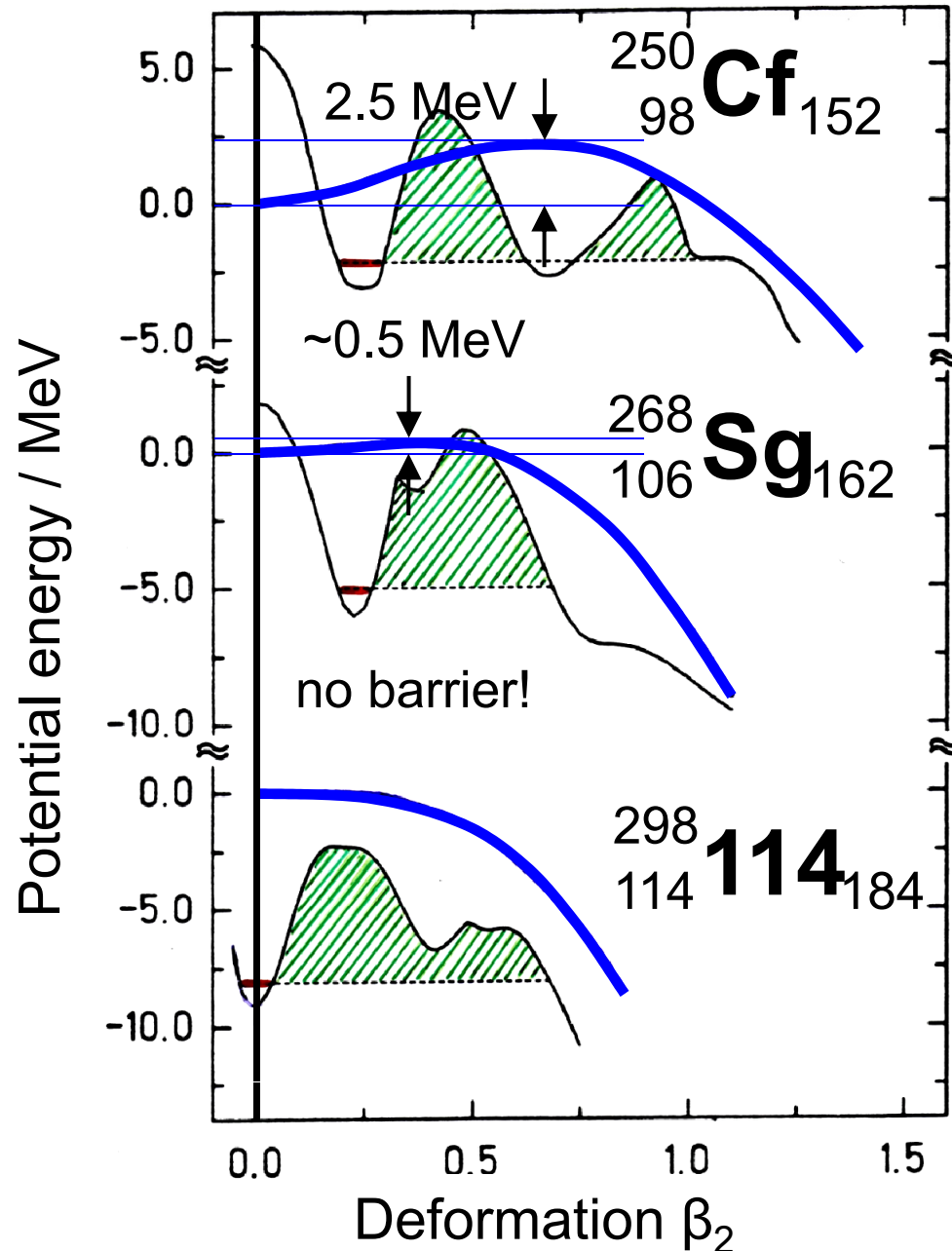


Influence of shell effects on fission barrier



— macroscopic barrier:
Disappears at $Z \sim 104$!

Influence of shell effects on fission barrier



— macroscopic barrier:

Disappears at $Z \sim 104$!

— with shell structure:

spherical \leftrightarrow deformed ground state

fission barrier is also > 0 for elements with $Z \geq 104$.

some fission barriers have complicated shapes, multiply-humped \Rightarrow fission isomers!

elements that exist only thanks to shell effects:

superheavy elements

Influence of Shell Effects on SF Half-Lives

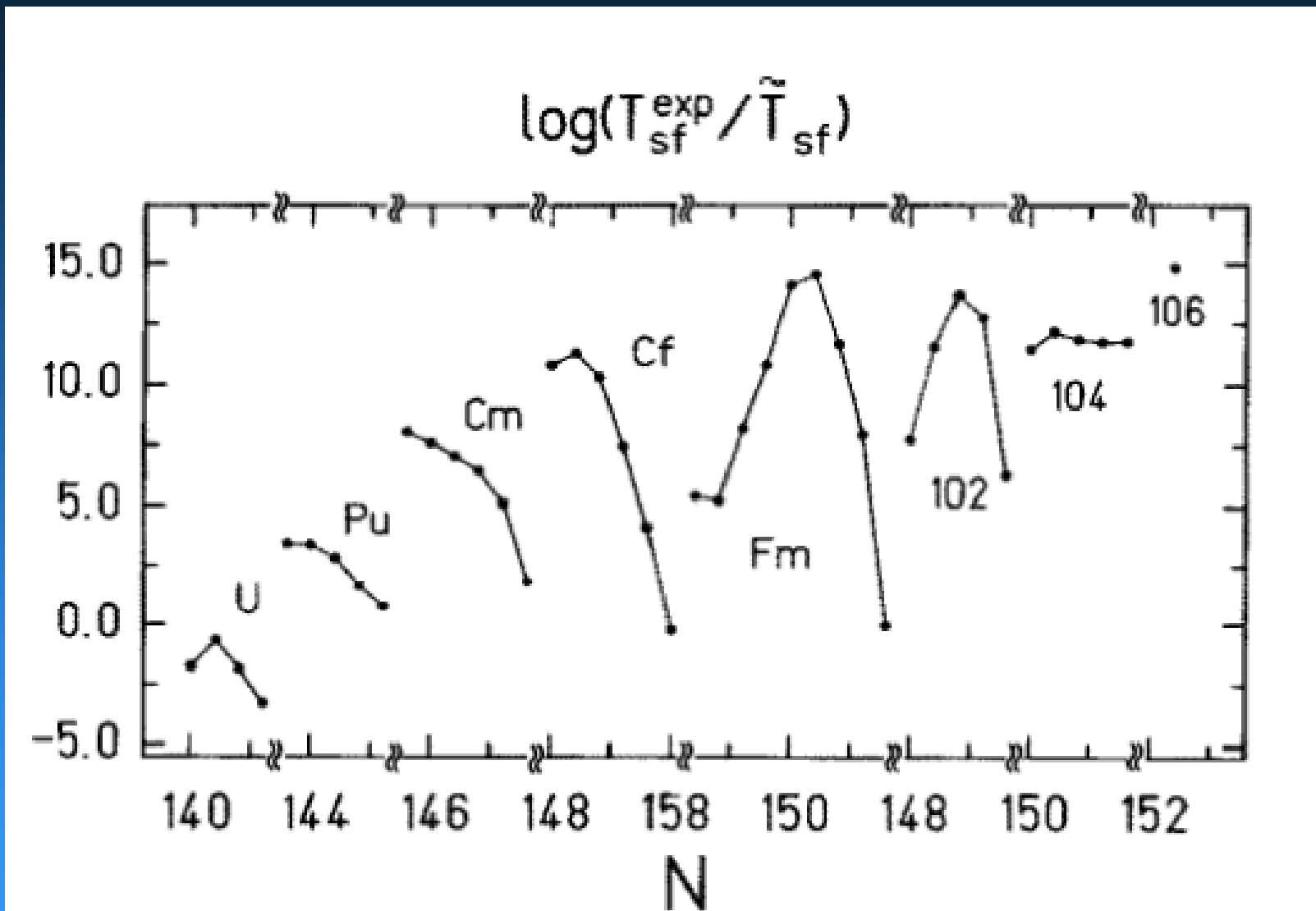
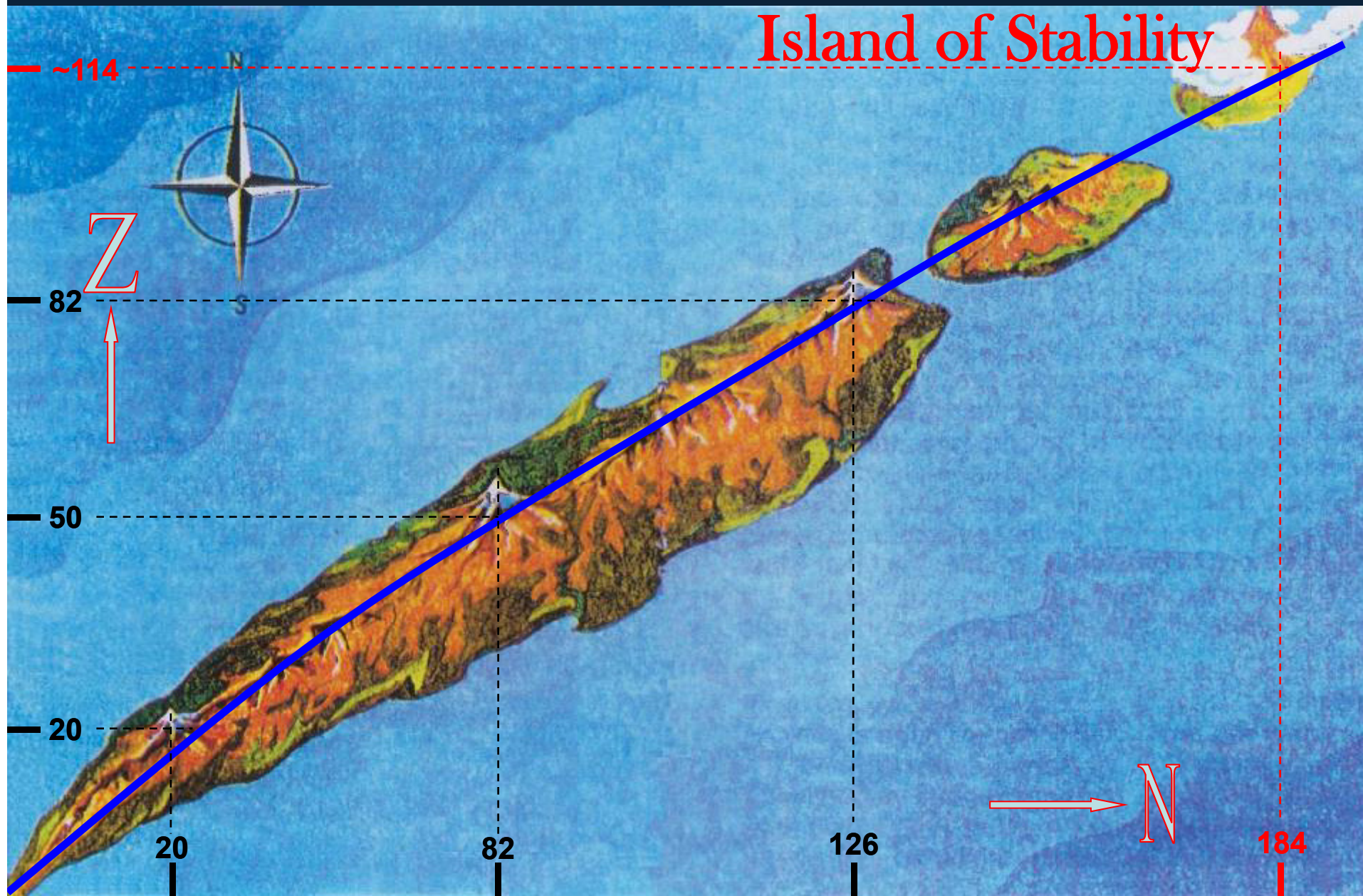
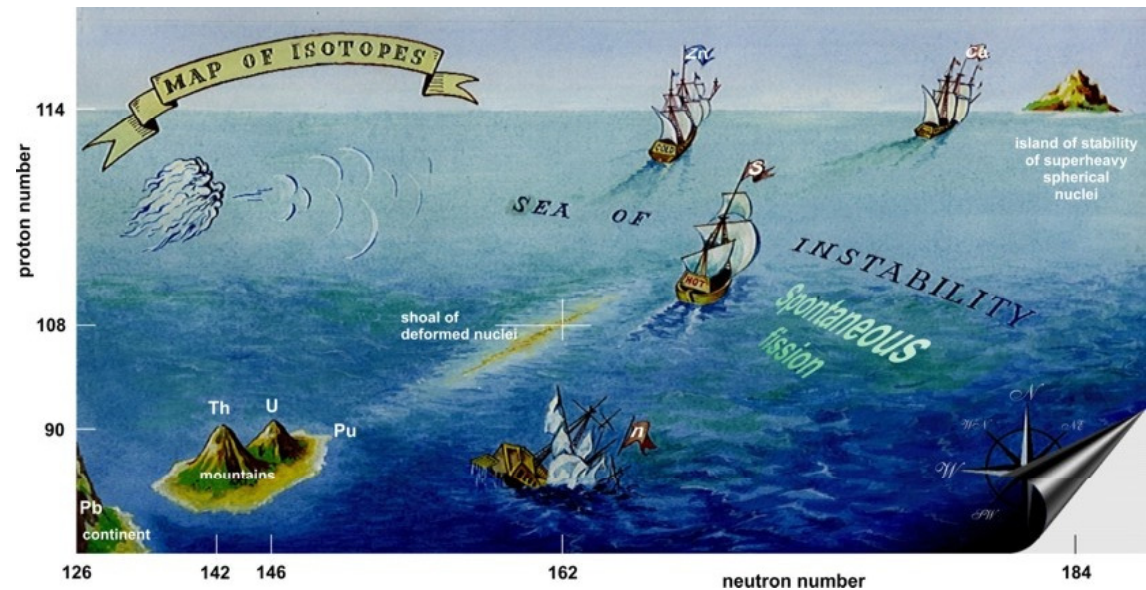
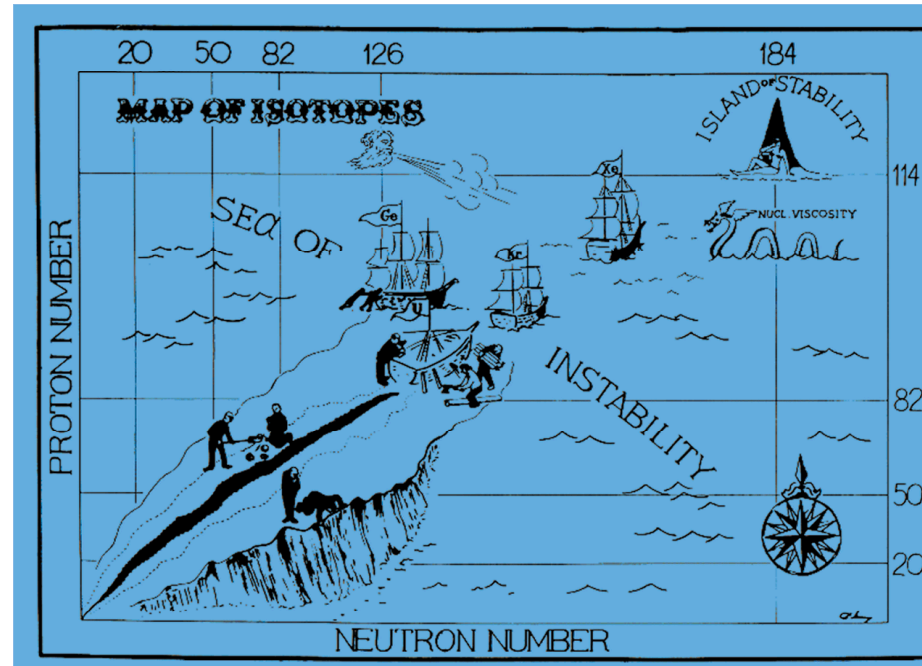
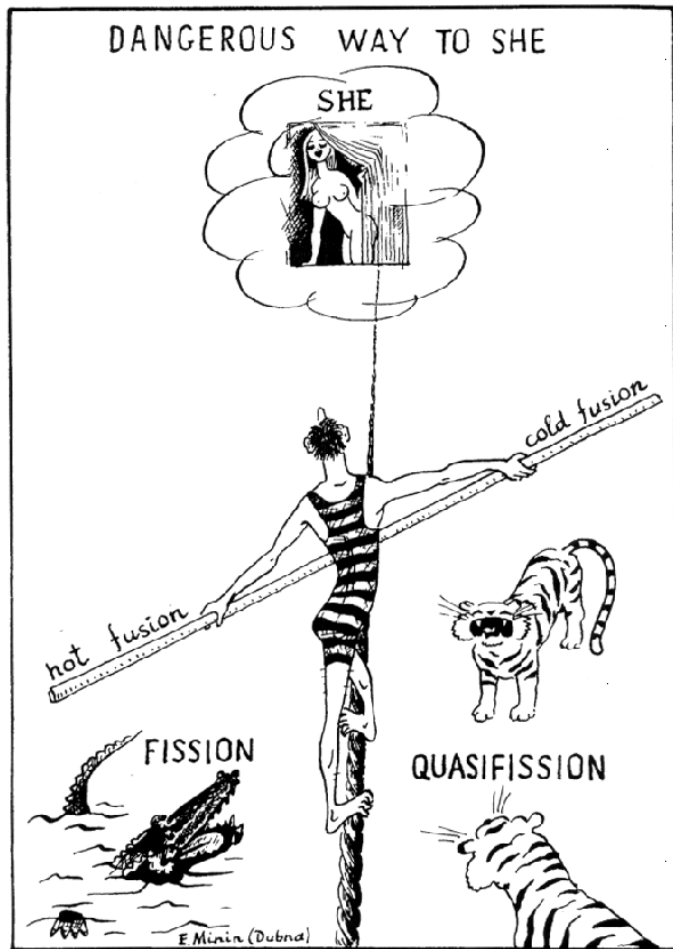


Fig. 8. Shell effect in the spontaneous-fission half-life T_{sf} .

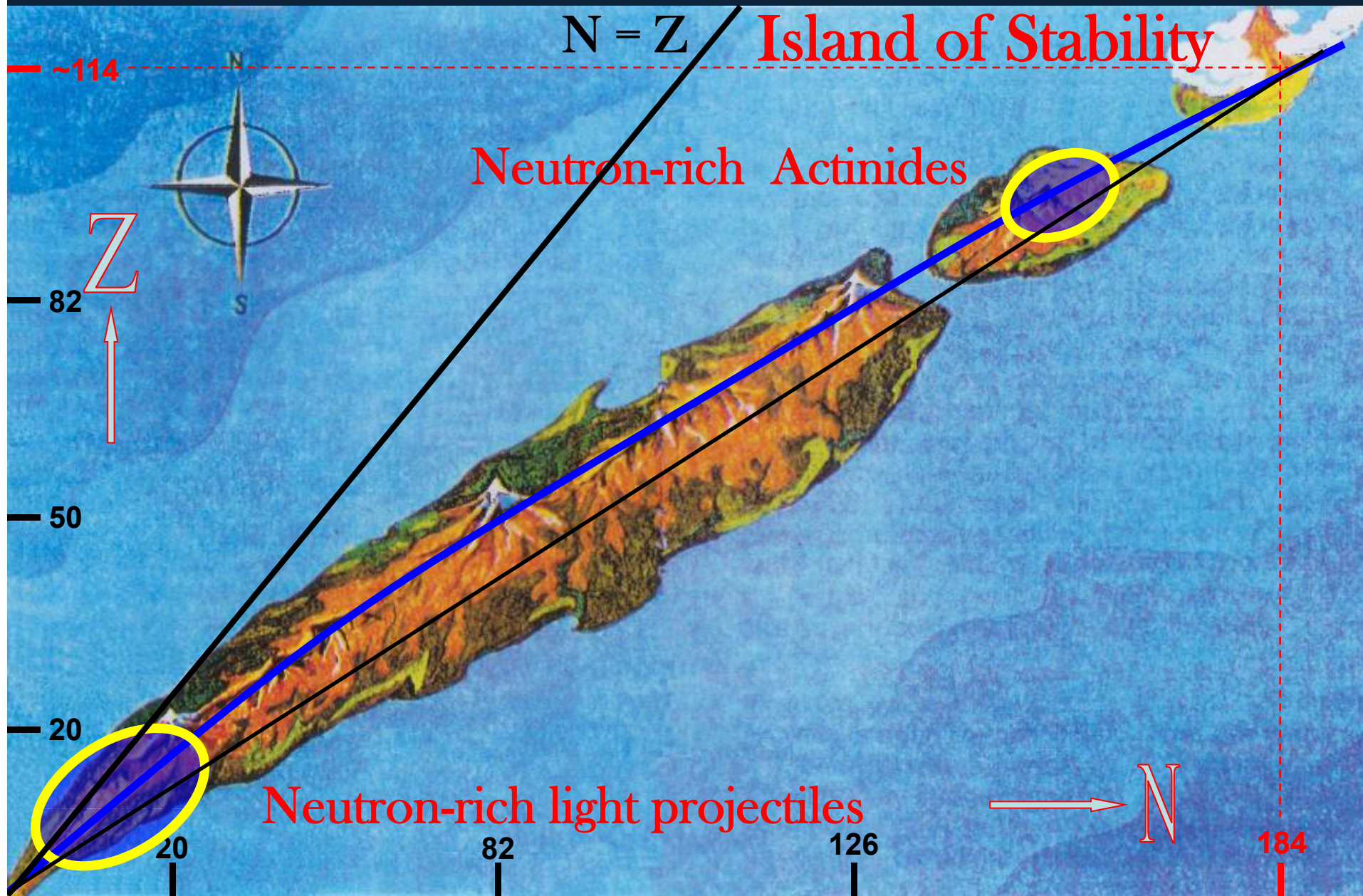
Map of the Nuclear Landscape



The quest for "SuperHeavy Elements" (SHE)



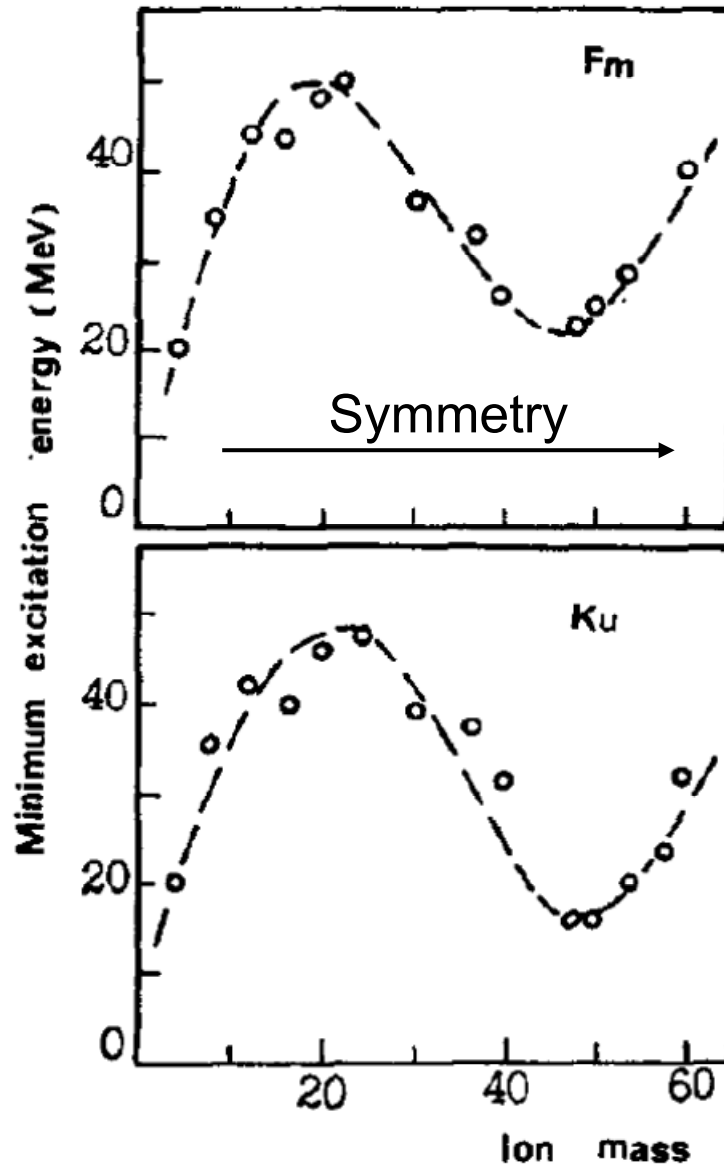
Map of the Nuclear Landscape



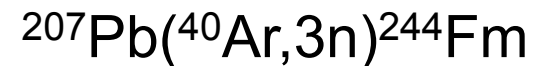
Md: Es target: 1e9 atoms
First "single atom chemistry"



FLNR/GSI: Cold fusion



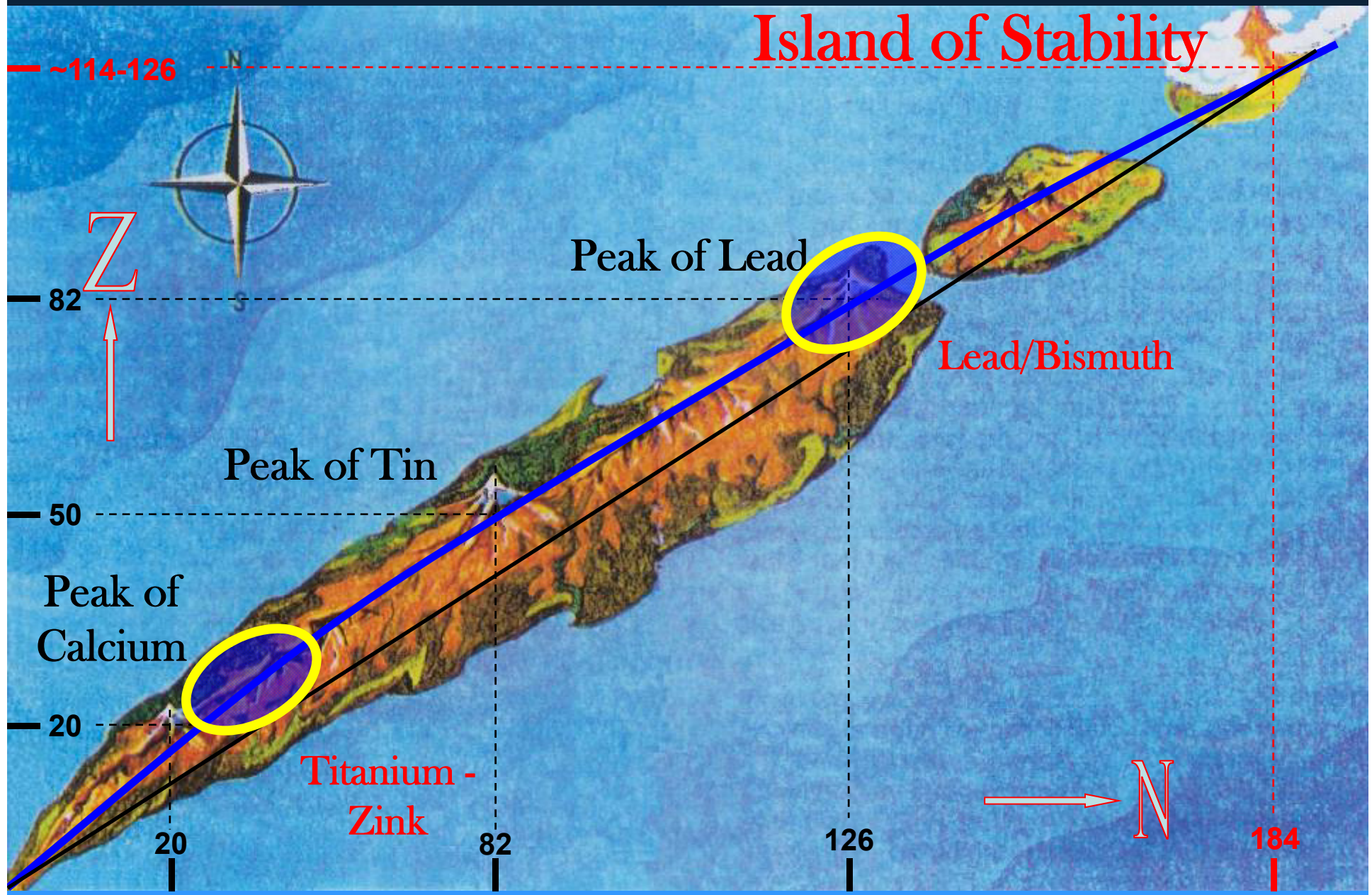
Shell effects around ^{208}Pb
decrease excitation energy in CN
→ evaporation of fewer neutrons
→ loss in exit channel smaller
→ should enhance cross sections



→ high cross sections measured at
FLNR!

Fig. 5. Minimum excitation energy of the compound nuclei ^{248}Fm and ^{258}Ku formed in different target-projectile combinations. The dashed curves are drawn through the calculated E^*_{min} values shown by points.

Map of the Nuclear Landscape

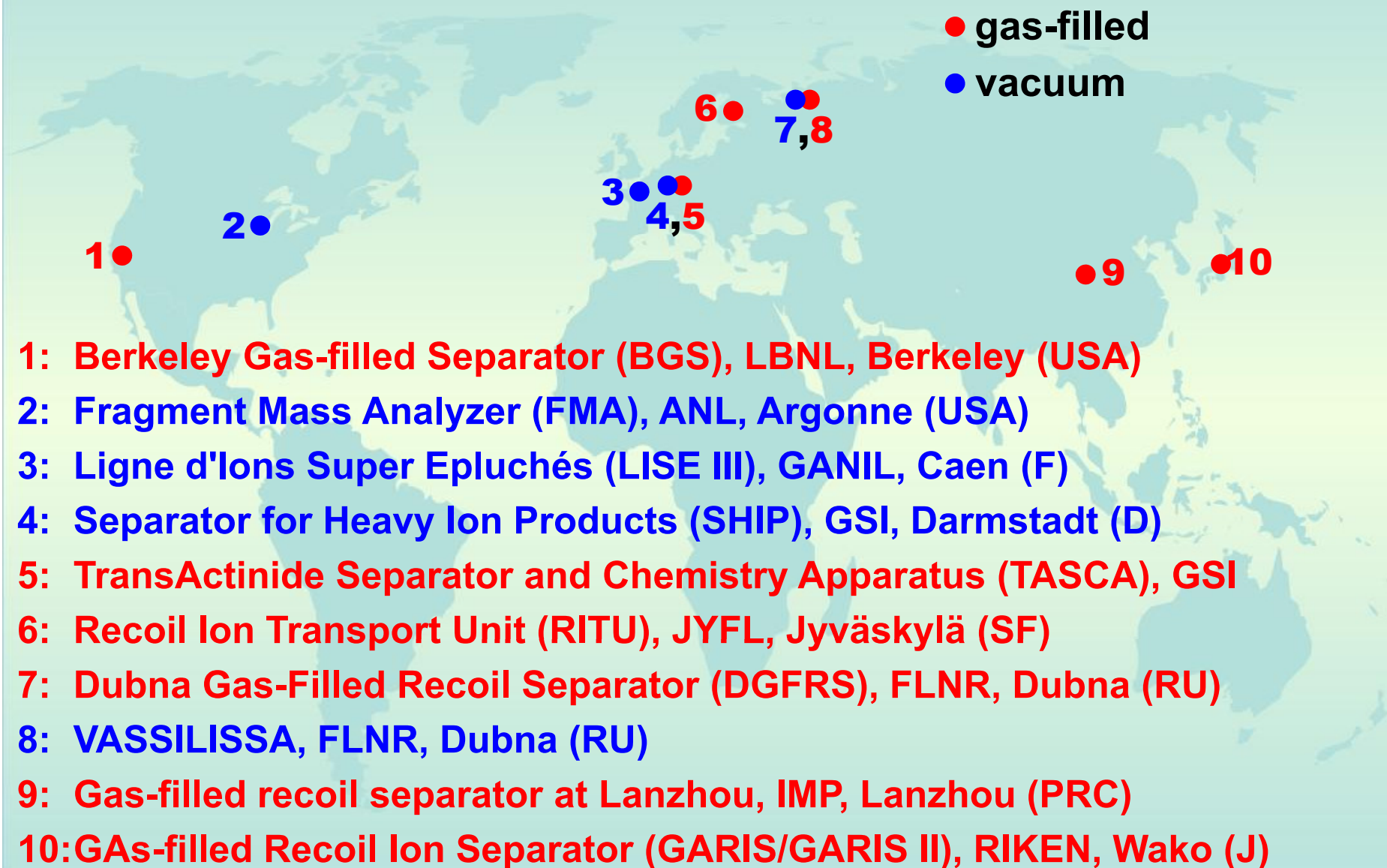


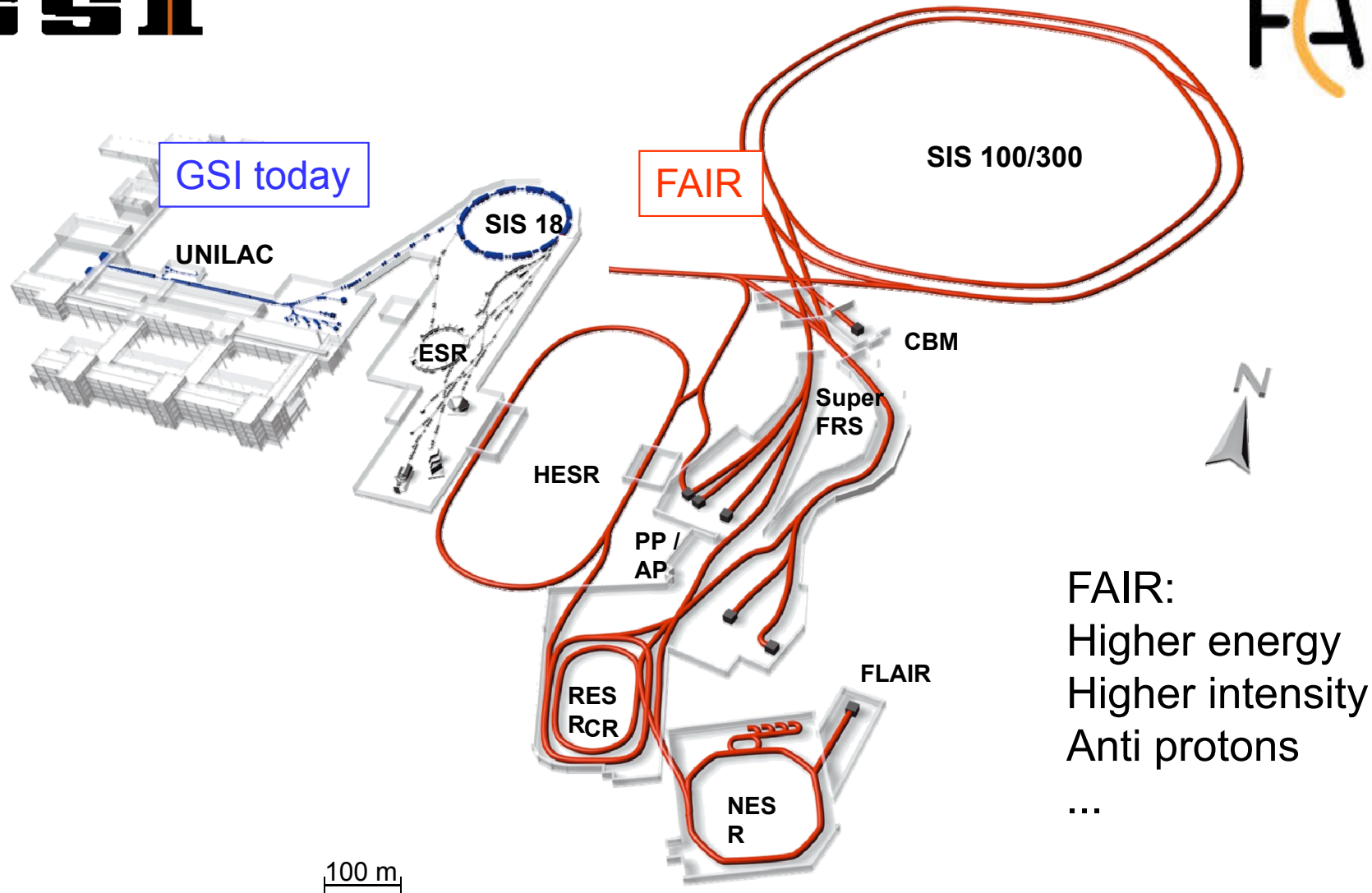
Experimental Techniques

Introduction to GSI

**Physics experiments at
recoil separators**

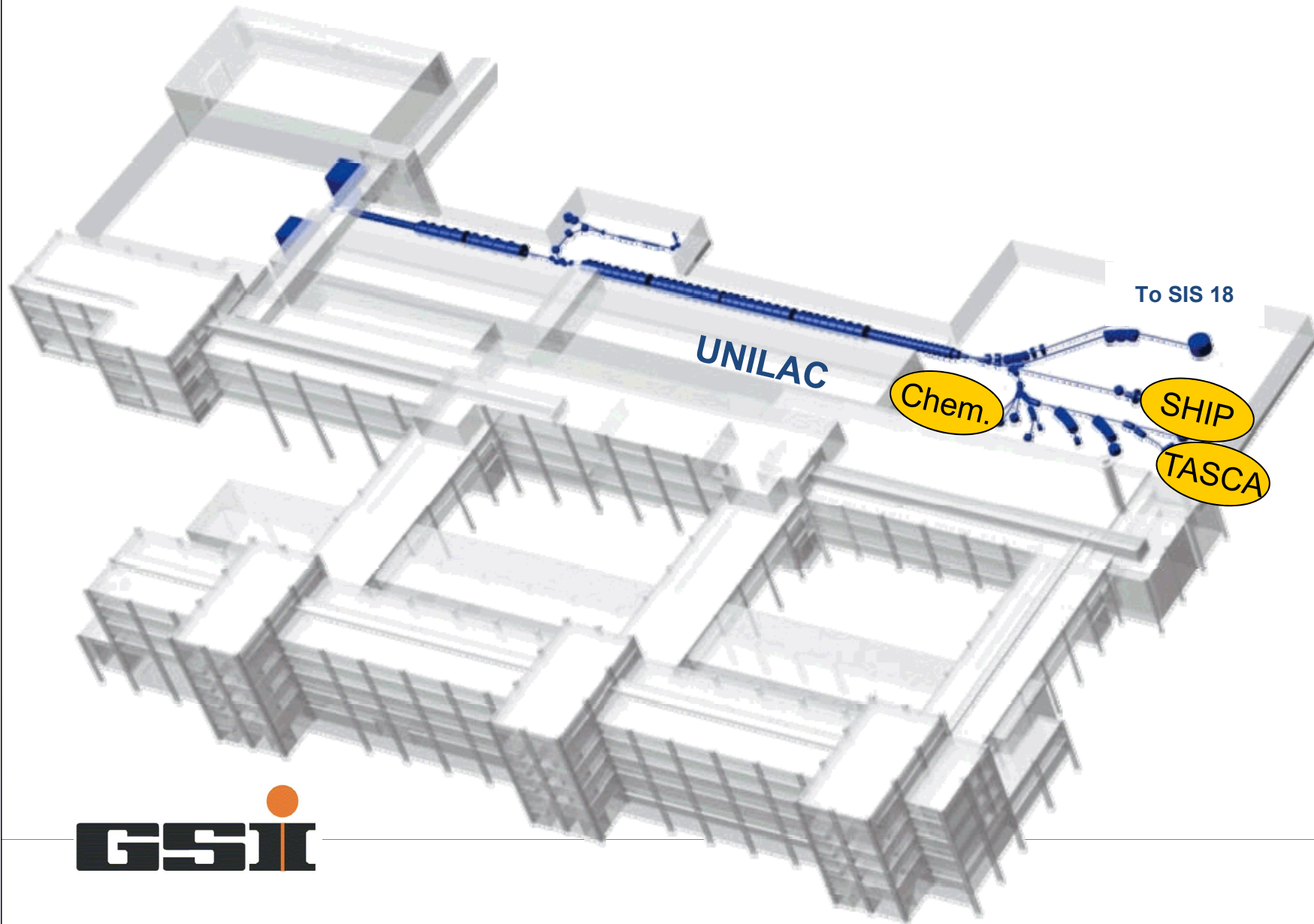
SHE Separators world-wide



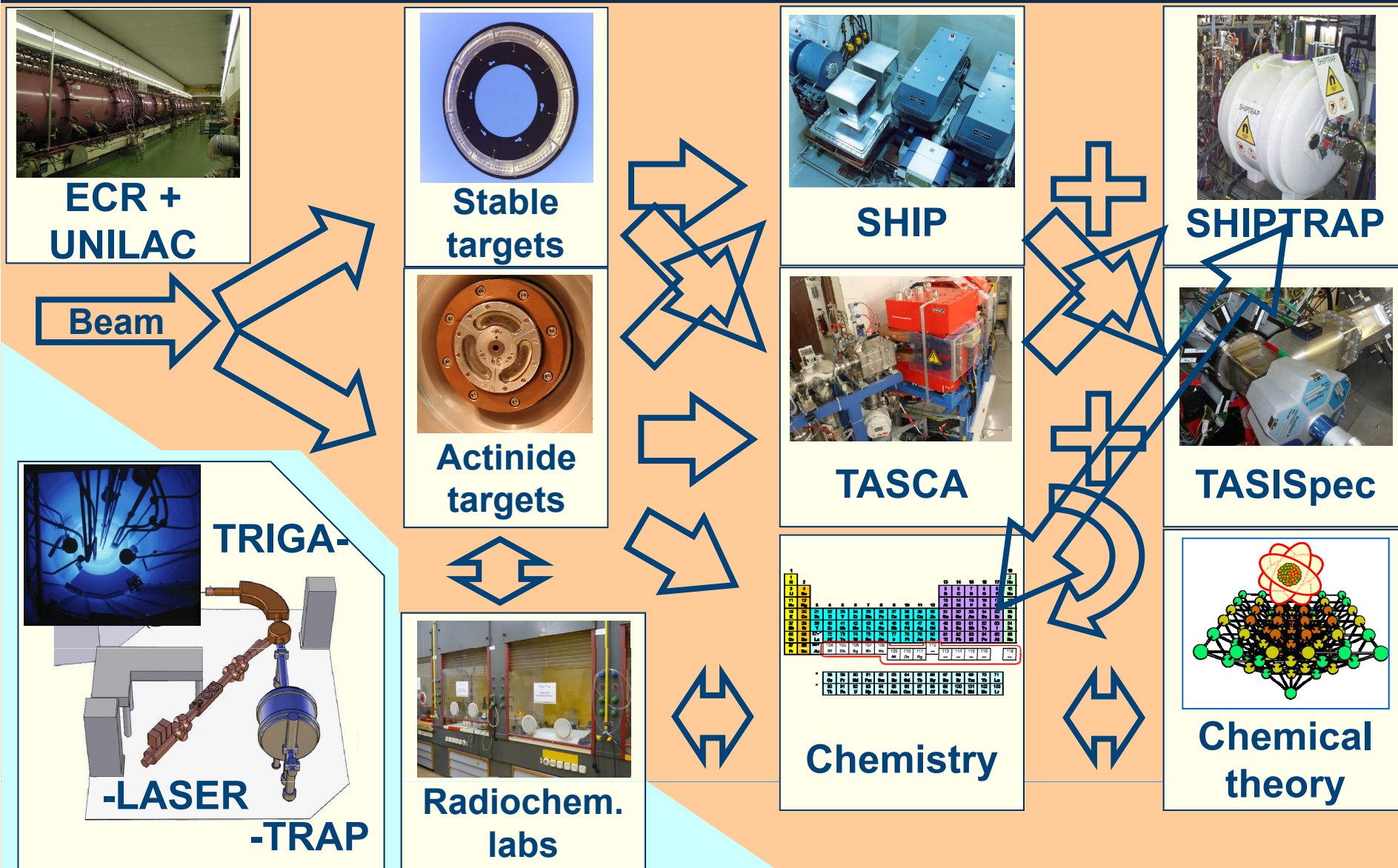


GSI Helmholtzzentrum für Schwerionenforschung mbH Facility for Antiproton and Ion Research

SHE Research at GSI



Unique Combination for SHE



Separation in E and M Dipol

Different ionic species are spatially separated by exploiting their different response (=radius of curvature) to the Lorentz force:

$$\vec{\mathbf{F}} = \mathbf{q} \cdot \left(\vec{\mathbf{E}} + \vec{\mathbf{v}} \times \vec{\mathbf{B}} \right)$$

Separators are built such that all vectors are orthogonal (or radial, in case of the E dipole). Then equating with the centrifugal force yields the respective rigidities:

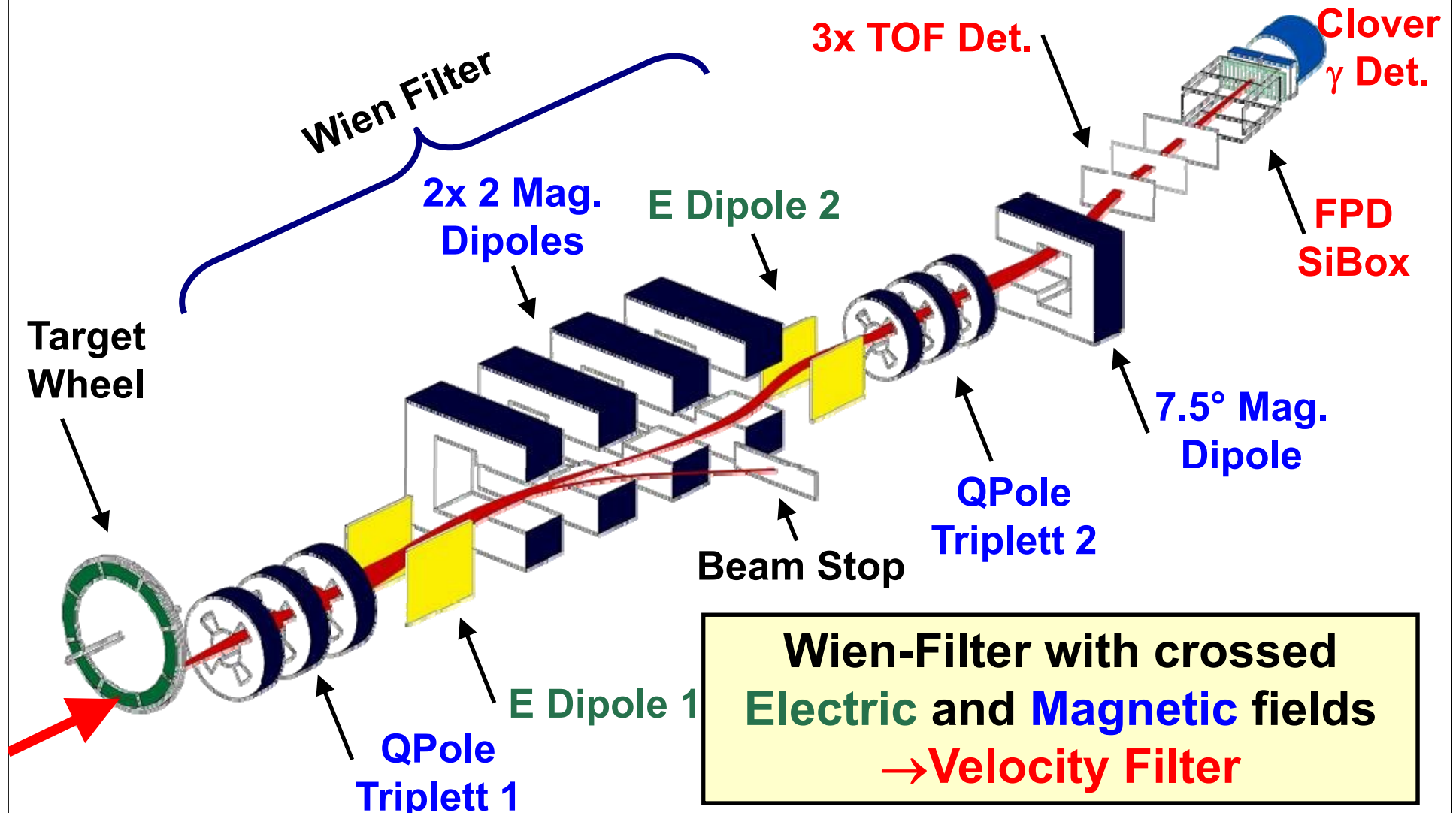
$$\text{Electric dipole:} \quad \mathbf{E} \cdot \rho \text{ [MV]} = \frac{\mathbf{m} \cdot \mathbf{v}^2}{\mathbf{q}}$$

$$\text{Magnetic dipole:} \quad \mathbf{B} \cdot \rho \text{ [T} \cdot \mathbf{m]} = \frac{\mathbf{m} \cdot \mathbf{v}}{\mathbf{q}}$$

Magnetic quadrupole multiplets as ion-optical lenses for focusing

Separation in Vacuum

Separator for Heavy Ion reaction Products – SHIP

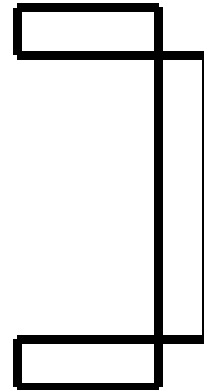


Generic Recoil Separator Detection System

TOF
Start



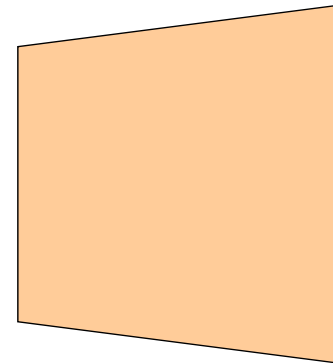
Focal
Plane +
Upstream



Punch
thru
Veto



Gamma /
X-Ray
Detector



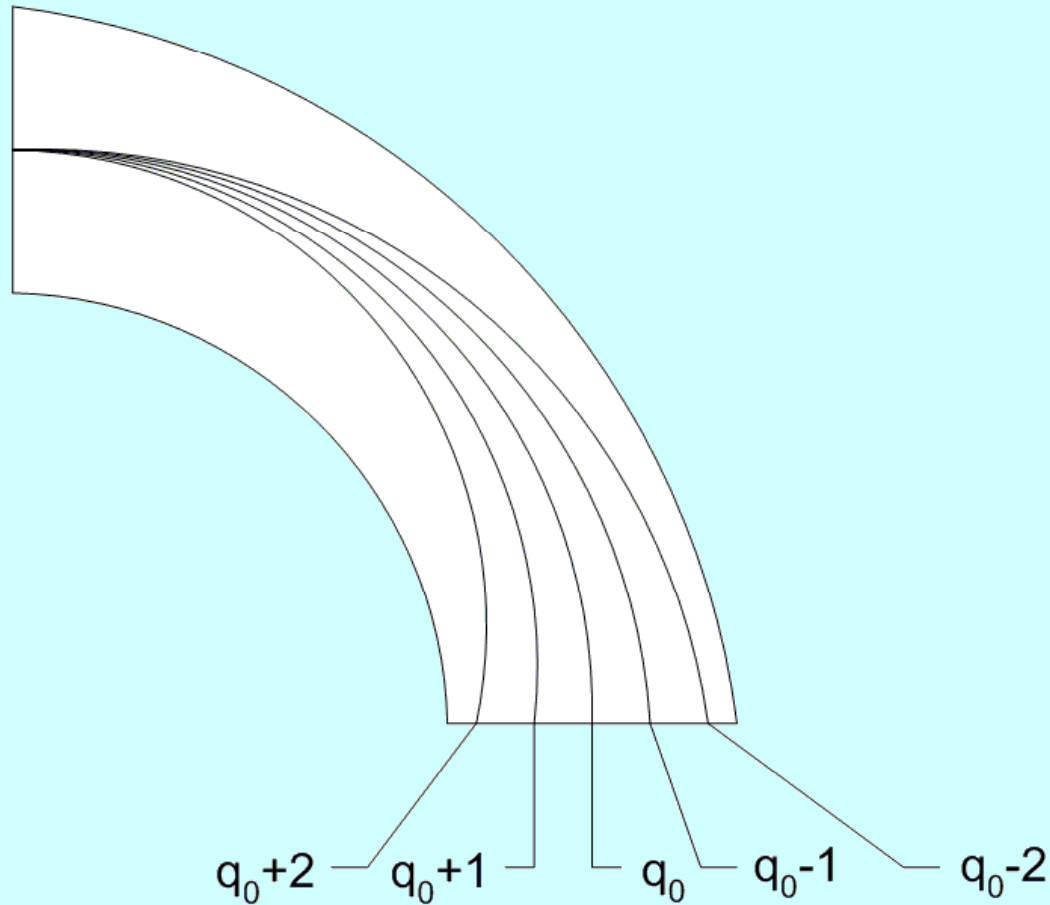
FPD Registers:

α -particles

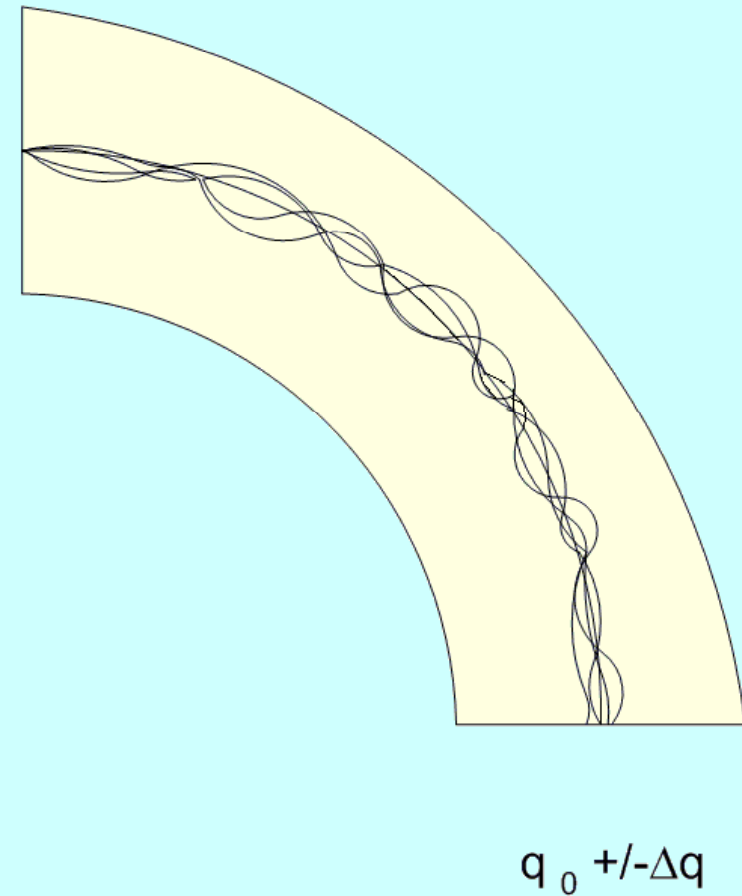
Conversion Electrons

Separation in gas-filled dipoles

In vacuum



In dilute gas



TASCA



www.gsi.de/tasca

TASCA

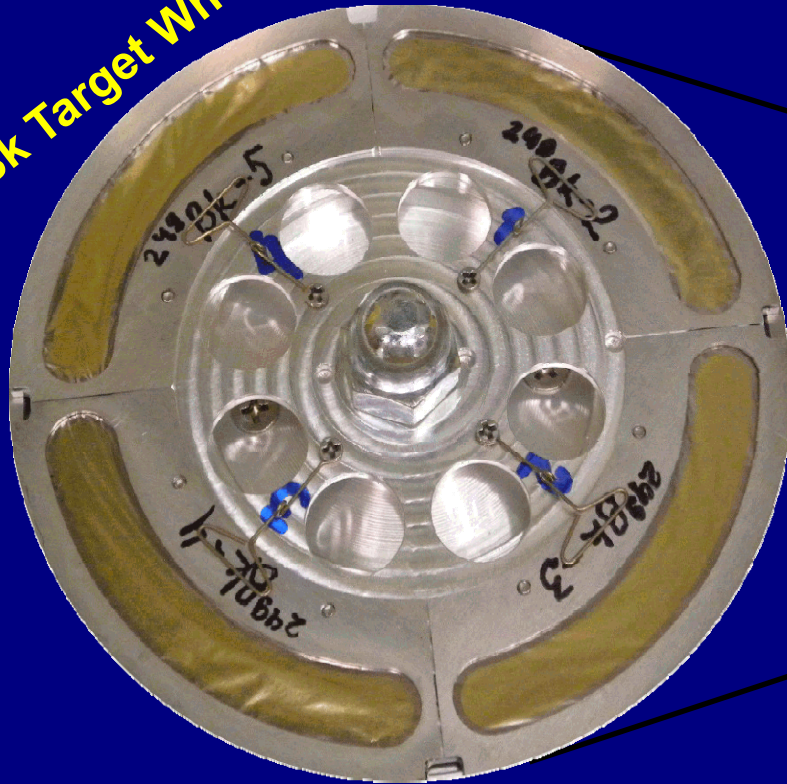
Target Chamber
side view

^{50}Ti beam

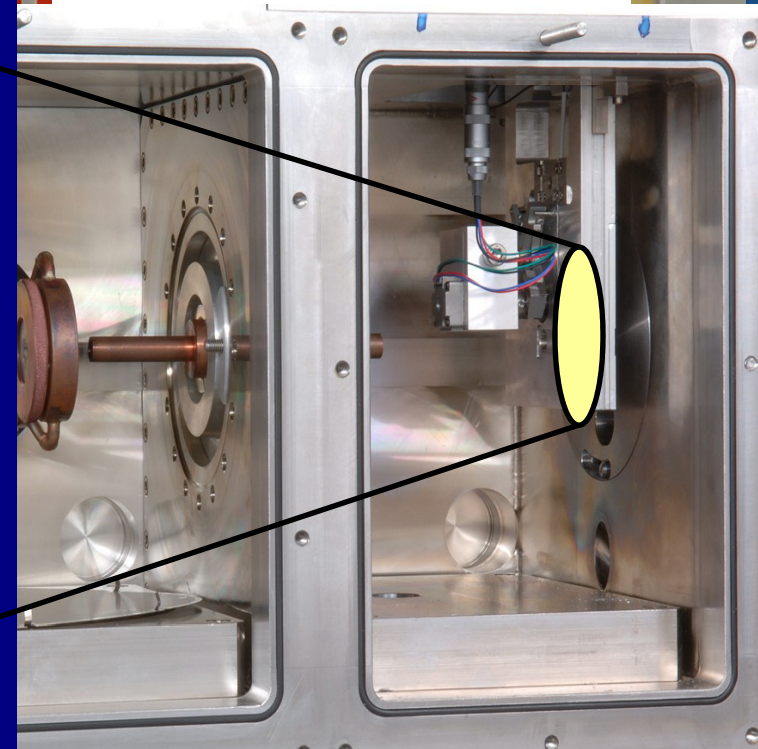
www.gsi.de/tasca

TASCA

249Bk Target Wheel



**Target Chamber
side view**



www.gsi.de/tasca

TASCA



DQQ-configuration
 $B\rho_{\max} \approx 2.4 \text{ Tm}$

www.gsi.de/tasca

TAS

DSSSD State-of-the-Art Stop Detector Array

6900 pixels

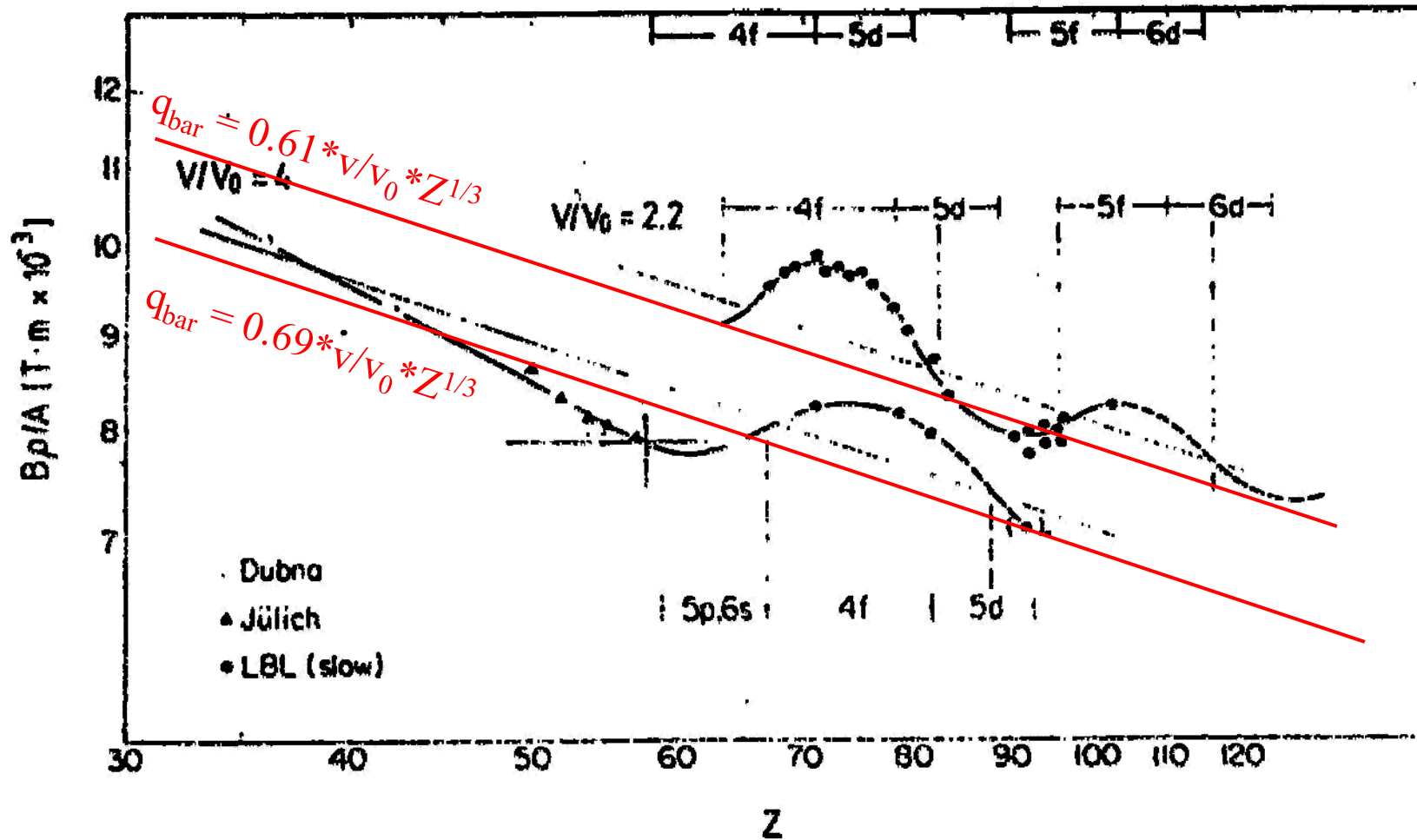
144 mm

48 mm

www.gsi.de/tasca

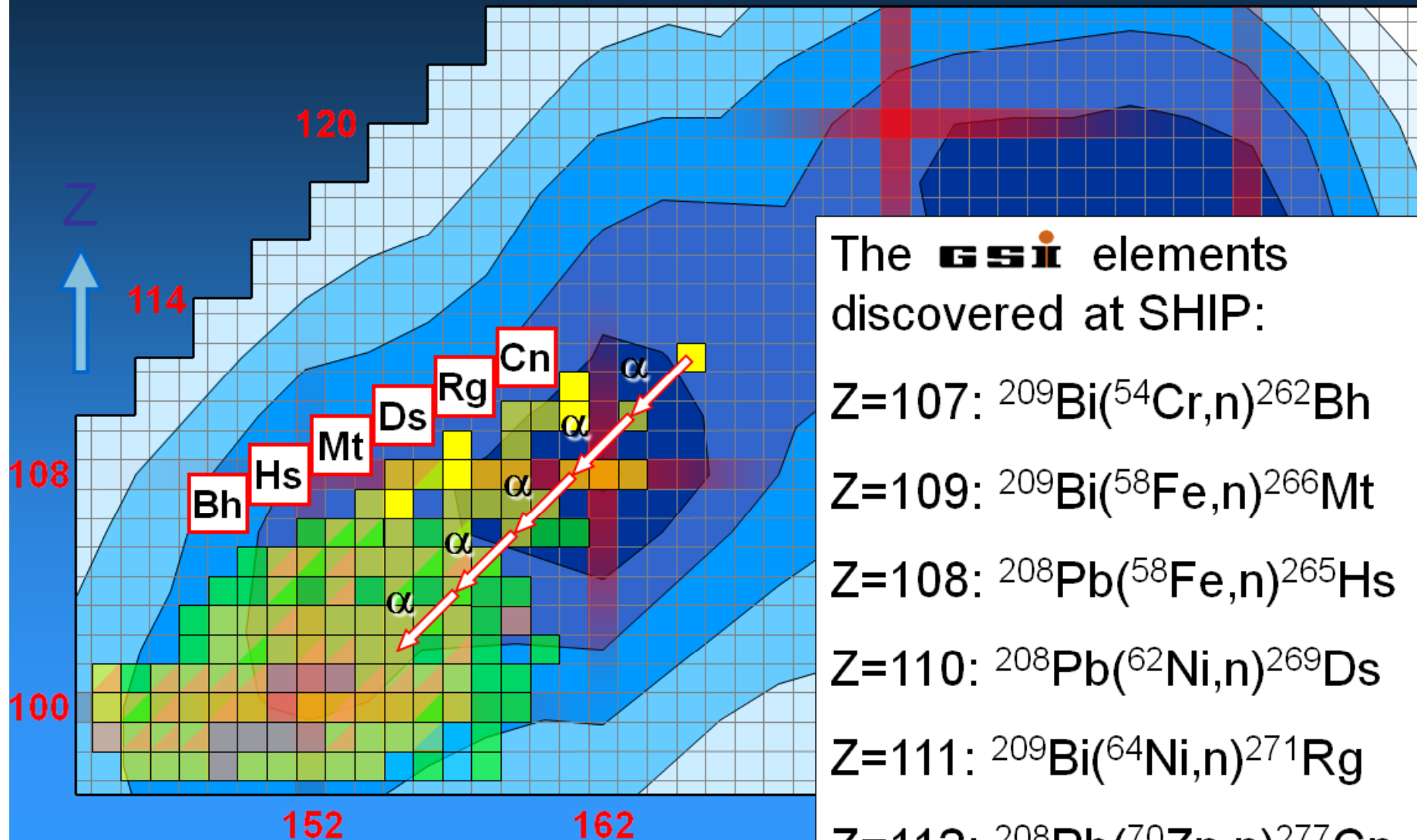
Average charge state in dilute He

$$B\rho \text{ [T}\cdot\text{m]} = m\cdot v/q \approx 0.0227\cdot(v/v_0)\cdot A/q$$

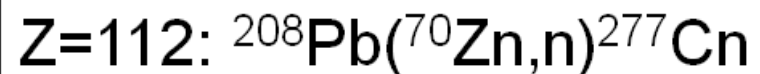
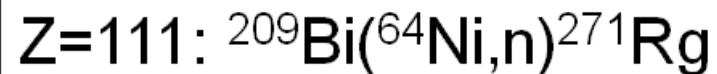
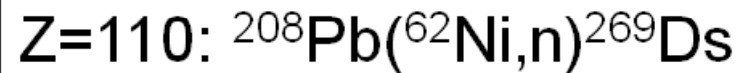
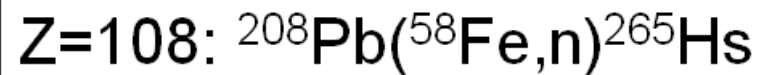
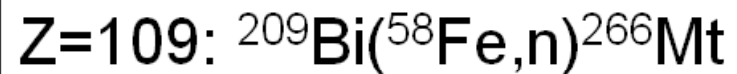
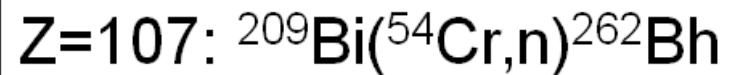


Ghiorso et al., Nucl. Instrum. Meth. A 269, 192 (1988)

Superheavy elements on the mic-mac landscape of today



The **GSI** elements discovered at SHIP:



SHIP Excitation Functions with ^{208}Pb Targets

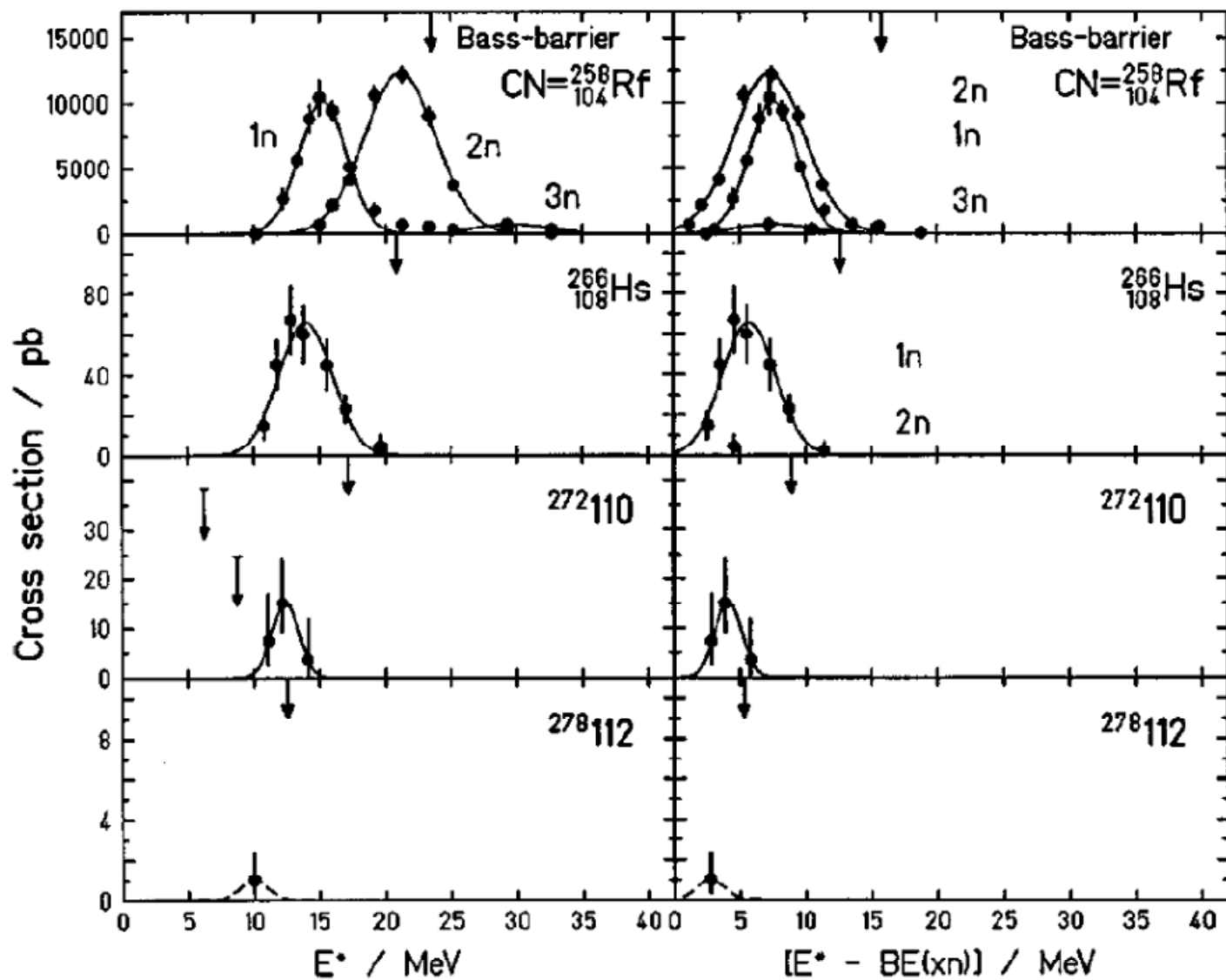
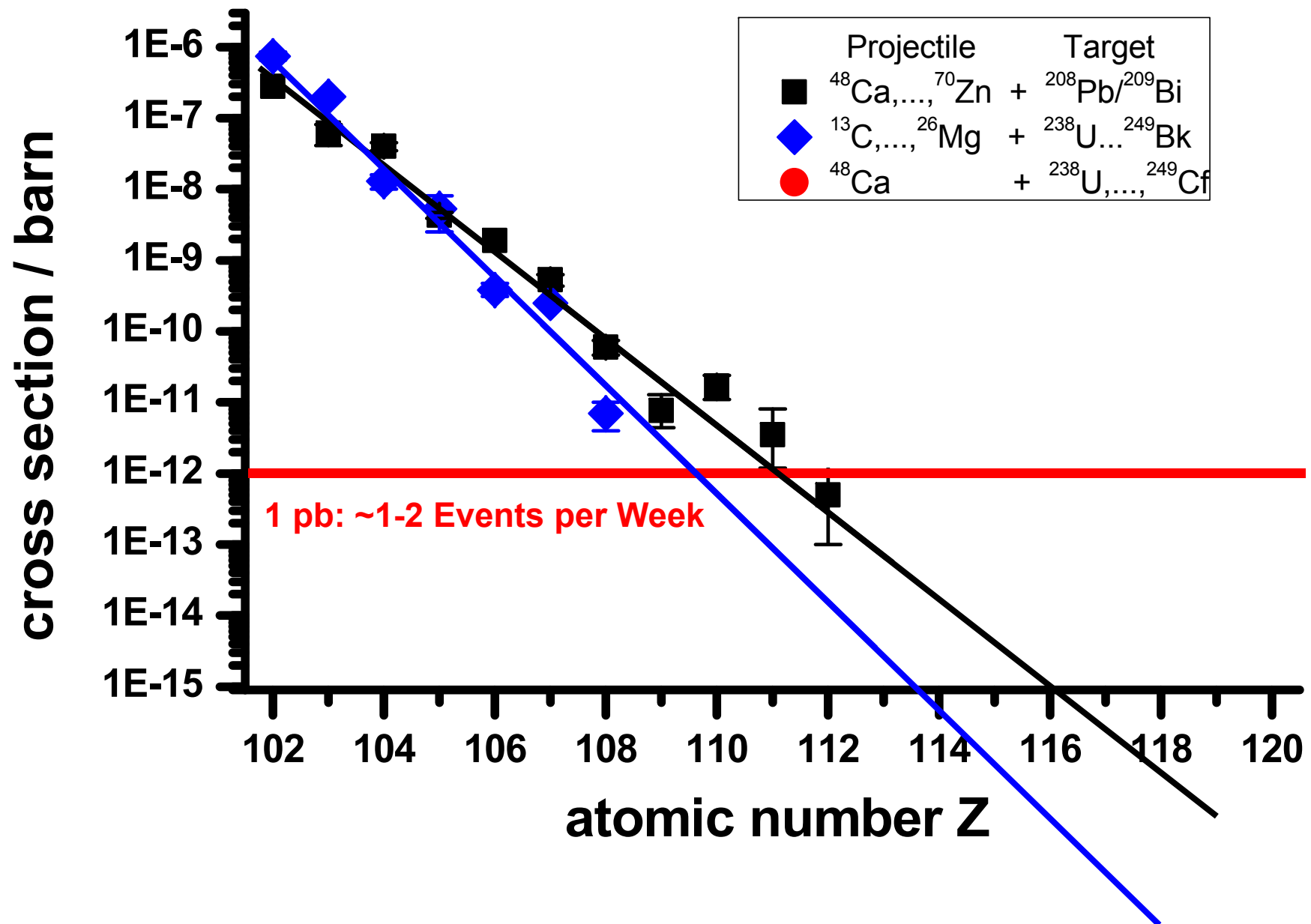


FIG. 18. Measured even-element excitation functions.

S. Hofmann et al.

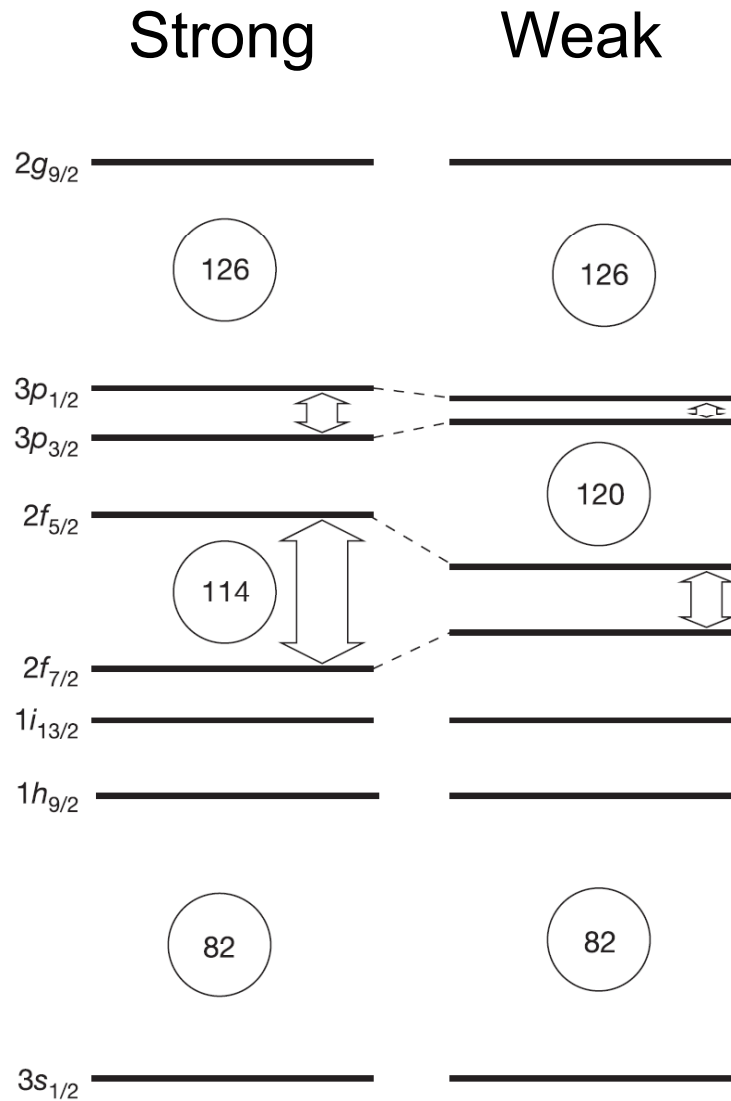
Cross Sections in Hot / Cold / ^{48}Ca Induced Fusion Reactions



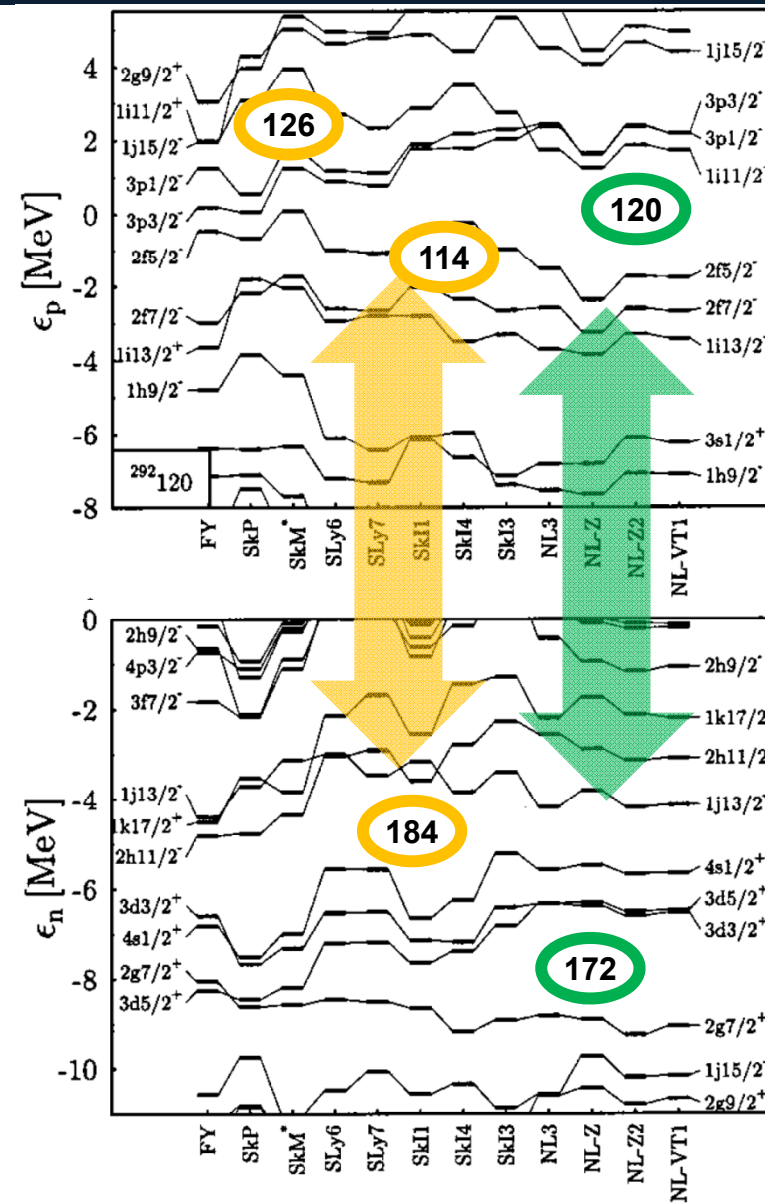
Question

Now where is the next magic number beyond $Z=82$?

Spin orbit splitting in superheavy nuclei

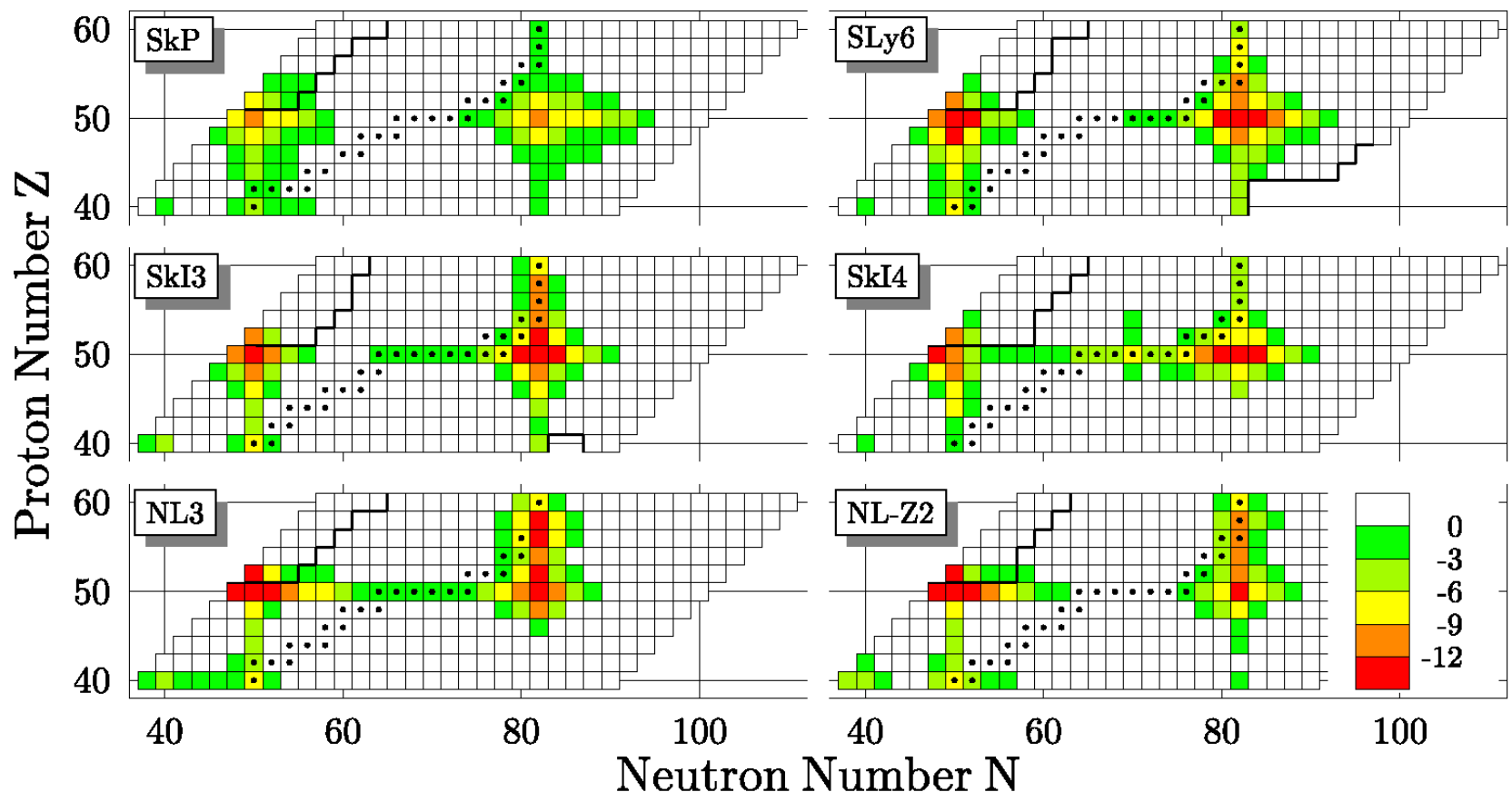


R.-D. Herzberg et al., Nature 442 (2006) 896



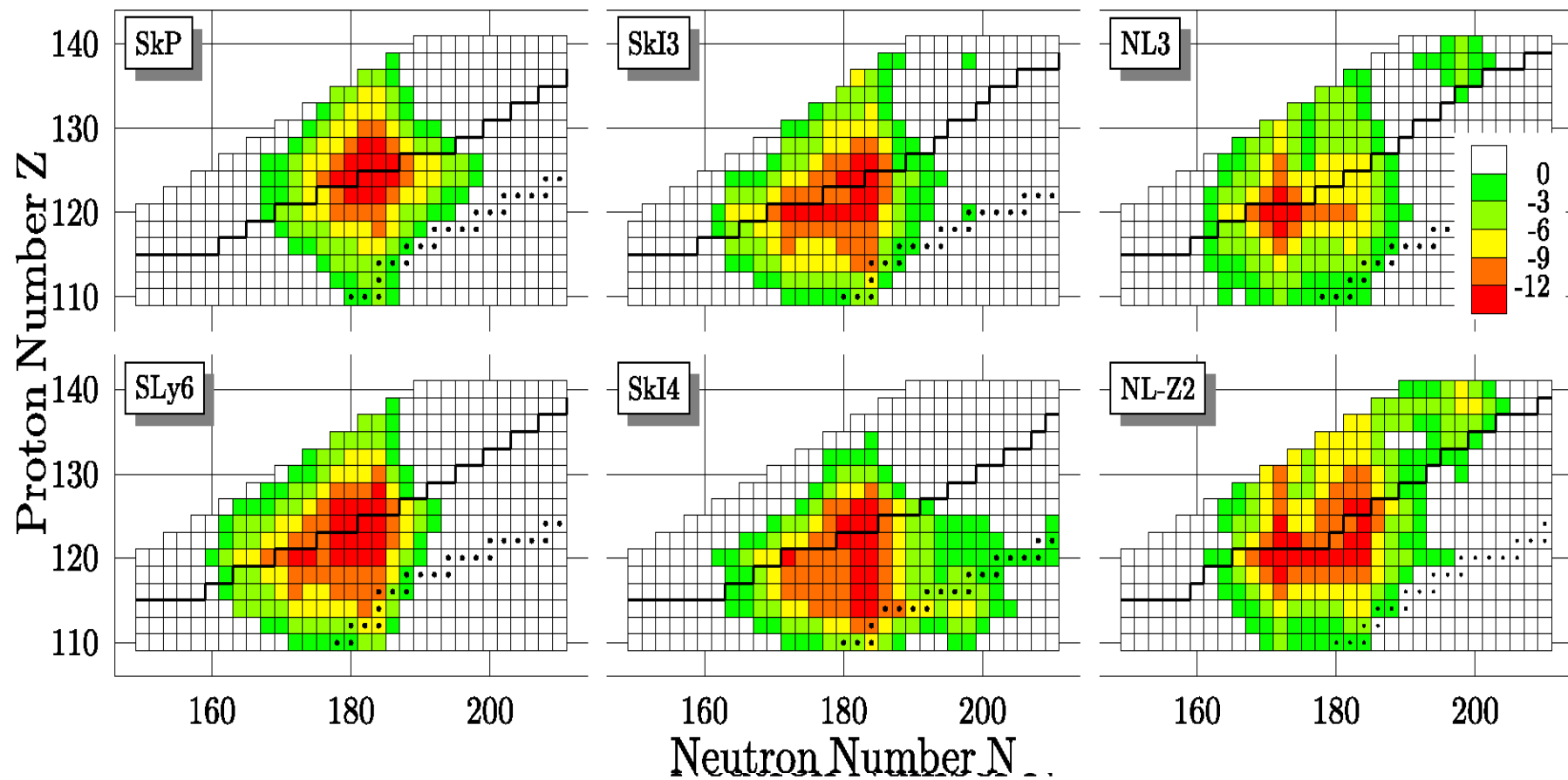
M. Bender et al., Phys. Rev. C 60 (1999) 034304

Nuclear shells



M. Bender et al., Phys. Lett. B 515 (2001) 42

Nuclear shells



M. Bender et al., Phys. Lett. B 515 (2001) 42